# The use of $\pi$-allyltricarbonyliron lactone complexes in the synthesis 

 of the resorcylic macrolides $\alpha$ - and $\boldsymbol{\beta}$-zearalenolSvenja Burckhardt and Steven V. Ley *<br>Department of Chemistry, University of Cambridge, Lensfield Rd, Cambridge, UK CB2 1EW

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A highly stereoselective synthesis of the natural products $\alpha$ - and $\beta$-zearalenol $\mathbf{1}$ and $\mathbf{2}$ has been achieved using $\pi$-allyltricarbonyliron lactone complexes to control the 1,5 -stereochemical relationship of the oxygen functionalities found in these resorcylic macrolides.

## Introduction

The 14 -membered resorcylic macrolides $\alpha$ - and $\beta$-zearalenol $\mathbf{1}$ and $\mathbf{2}$ are estrogenic mycotoxins produced by a species of the fungus Fusarium. ${ }^{1}$ The zearalenols 1 and 2, together with the reduced isomer $\alpha$-zearalanol 3 , belong to a class of biologically important compounds whose progenitor is considered to be zearalenone 4 (Fig. 1). The hormonal activity of these

$\alpha$-zearalenol 1

$\alpha$-zearalanol 3

$\beta$-zearalenol 2

zearalenone 4

Fig. 1 The resorcylic acid lactones $\alpha$-zearalenol 1, $\beta$-zearalenol 2, $\alpha$-zearalanol 3 and zearalenone 4.
compounds is linked to their close spatial similarity to $17 \beta-$ estradiol, ${ }^{2}$ with the $\alpha$-isomer $\mathbf{1}$ being three to four times as active as the $\beta$-isomer $\mathbf{2}$. ${ }^{3}$ Furthermore this class of compounds has attracted attention due to their anabolic activity. For example, $\alpha$-zearalanol $\mathbf{4}$ is employed as a cattle-growth stimulant ${ }^{4}$ and also has undergone clinical trials as a potential treatment for menopausal and post-menopausal syndrome. ${ }^{5}$ While several total syntheses of zearalenone $\mathbf{4}$ have been accomplished over the last 30 years, ${ }^{6}$ to our knowledge no independent direct syntheses of $\mathbf{1}$ or $\mathbf{2}$ have been described. Here we report in full on the first enantioselective route to $\mathbf{1}$ and $\mathbf{2}$ employing $\pi$-allyltricarbonyliron lactone complexes as unusual precursor molecules to these natural products. ${ }^{7}$

We have shown previously that organoaluminium reagents, possessing an active $\beta$-hydrogen atom, such as tripro-pyl- or triisobutylaluminium, will reduce carbonyl groups appended to the allyl ligand of $\pi$-allyltricarbonyliron lactone complexes with excellent diastereoselectivity. ${ }^{8}$ We have also shown that sodium triacetoxyborohydride efficiently decomplexes $\pi$-allyltricarbonyliron lactone complexes bearing a hydroxy group in the side-chain to generate stereodefined

1,5 -diols. ${ }^{9}$ Here we exploit these processes to show that $\pi$-allyltricarbonyliron lactone complexes can be used to control the required stereochemical arrangement found in $\mathbf{1}$ and $\mathbf{2}$.

## Results and discussion

The synthesis begins with the reduction of the ester $5^{10}$ to the corresponding alcohol, followed by Swern oxidation and Horner-Wadsworth-Emmons homologation using the phosphonate 6, prepared according to the method of Grieco and Pogonowski, ${ }^{11}$ providing the ( $E$ )-enone 7 in $83 \%$ yield over three steps (Scheme 1). Deprotection of the acetonide under acidic conditions and transformation of the liberated diol $\mathbf{8}$ to the cyclic sulfite 9 using thionyl chloride ${ }^{12}$ afforded 9 as a mixture of diastereoisomers in $82 \%$ overall yield. Treatment of 9 with nonacarbonyldiiron in benzene under sonication conditions ${ }^{13}$ provided the two diastereoisomeric $\pi$-allyltricarbonyliron lactone complexes, endo-10a and exo-10b, in 70\% combined yield and in a ratio of $c a .1: 1$. Separation of the two isomers 10a and 10b was readily achieved by flash column chromatography. This $1: 1$ mixture was ideal since it should be possible to process these isomers individually to synthesise selectively the two natural products $\mathbf{1}$ and 2. For example, for $\alpha$-zearalenol 1, reduction of the side-chain ketone in the endo complex 10a was achieved in $94 \%$ yield using tripropylaluminium ${ }^{8}$ to give 11a, as the sole product as determined by $600 \mathrm{MHz}{ }^{1} \mathrm{H}$ NMR analysis. Analogously, the exo complex 10b provided diastereomerically pure 11b in $80 \%$ yield for the synthesis of $\beta$-zearalenol 2. Treatment of 11a and 11b with sodium triacetoxyborohydride in tetrahydrofuran ${ }^{9}$ resulted in a highly stereoselective decomplexation to afford, after TBDMSprotection and hydrogenation, the alcohols 12a and 12b in 61 and $66 \%$ yield over three steps, respectively (Scheme 1). Next, Swern oxidation of 12a and 12b provided the corresponding aldehydes 13a and 13b which in turn were transformed into the vinylstannanes 14a and 14b by applying the procedure developed by Hodgson et al. ${ }^{14}$ utilising chromium(II) chloride and $\mathrm{Bu}_{3} \mathrm{SnCHI}_{2}$ in dimethylformamide (Scheme 2). With these stannanes in hand, we then examined their palladium catalysed coupling to the aromatic iodide $\mathbf{1 5}$, previously synthesised by Hegedus et al. ${ }^{6{ }^{6}}$ It soon became apparent that the nature of the ligand played a critical role in this Stille coupling reaction. The optimised conditions required the use of tetrakis(tri-2-furylphosphine)palladium, ${ }^{15}$ which was prepared in situ, to provide independently the coupled products 16a and 16b in 82 and $85 \%$ yields, respectively. Treatment of $\mathbf{1 6 a}$ or $\mathbf{1 6 b}$ with HF-pyridine followed by hydrolysis of the methyl ester functionality using aqueous potassium hydroxide in ethane-1,2-diol at $120{ }^{\circ} \mathrm{C}$


Scheme 1 Reagents and conditions: i. $\mathrm{LiAlH}_{4}, \mathrm{Et}_{2} \mathrm{O}, 0{ }^{\circ} \mathrm{C}, 2 \mathrm{~h}$; ii. $(\mathrm{COCl})_{2}, \mathrm{DMSO}_{2}, \mathrm{Et}_{3} \mathrm{~N}, \mathrm{DCM},-78{ }^{\circ} \mathrm{C}, 3 \mathrm{~h}$; iii. ( EtO$)_{2} \mathrm{P}(\mathrm{O}) \mathrm{CH}_{2}$ $\mathrm{CO}\left(\mathrm{CH}_{2}\right)_{4} \mathrm{OBn} 6, \mathrm{NaH}, \mathrm{THF},-78{ }^{\circ} \mathrm{C}, 1 \mathrm{~h}, 83 \%$ (over 3 steps); iv $\mathrm{AcOH}-\mathrm{H}_{2} \mathrm{O}(1: 1), 40^{\circ} \mathrm{C}, 24 \mathrm{~h}, 92 \%$; v. $\mathrm{SOCl}_{2}, \mathrm{Et}_{3} \mathrm{~N}, \mathrm{Et}_{2} \mathrm{O}, 0^{\circ} \mathrm{C}, 30 \mathrm{~min}$, $89^{\%} \%$; vi. $\mathrm{Fe}_{2}(\mathrm{CO})_{9}$, benzene, sonication, $30^{\circ} \mathrm{C}, 3 \mathrm{~h}, 35 \% \mathbf{1 0 a}, 35 \% \mathbf{1 0 b}$; vii. $\mathrm{AlPr}^{\mathrm{n}}{ }_{3}, \mathrm{DCM}, 0^{\circ} \mathrm{C}, 94 \% 11 \mathrm{a}, 80 \% 11 \mathrm{~b}$; viii. $\mathrm{NaBH}(\mathrm{OAc})_{3}$, THF, 3 d ; ix. TBDMSCl, imidazole, DMF, $0^{\circ} \mathrm{C}, 30 \mathrm{~min}$, then rt, 24 h ; x. Pd/C $(10 \%), \mathrm{H}_{2}$, EtOAc, $30 \mathrm{~min}, 61 \%$ 12a (over three steps), $66 \%$ 12b (over three steps).
provided the corresponding seco acids 18a and 18b in 83 and $85 \%$ yield over two steps.

Finally we investigated methods for the macrolactonisation of these seco acids to the 14 -membered MEM-protected $\alpha$ - and $\beta$-zearalenone products 19a and 19b. Application of the CoreyNicolaou procedure, ${ }^{6 b, 16}$ involving the formation of pyridine-2thiol esters and thermal cyclisation, did not provide any of the cyclised products, while the use of the Yamaguchi macrolactonisation ${ }^{17}$ gave a low yielding and disappointing 1: 1 mixture of the $10-$ and 14 -membered lactones, arising from cyclisation of the acid functionality with the $\mathrm{C}-6^{\prime}$ or the $\mathrm{C}-10^{\prime}$ hydroxy group respectively. Pleasingly however, cyclisation of 18a using Mukaiyama's protocol ${ }^{18}$ afforded the desired MEMprotected $\alpha$-zearalenol 19a in $64 \%$ yield, while the corresponding MEM-protected $\beta$-zearalenol 19b was obtained in $62 \%$ yield from 18b. In both cases only very minor traces of the corresponding 10 -membered lactones were observed by 600 $\mathrm{MHz}{ }^{1} \mathrm{H}$ NMR analysis in the crude mixture. Final deprotection of the MEM-ethers in 19a and 19b proceeded in $93 \%$ yield using aqueous hydrochloric acid in tetrahydrofuran at $40^{\circ} \mathrm{C}$ to provide $\alpha$-zearalenol 1 and $\beta$-zearalenol 2, respectively (Scheme 2). The products were identical in all respects to authentic samples.
In summary, a synthetic route to the estrogenic mycotoxins $\alpha$ - and $\beta$-zearalenol has been achieved whereby $\pi$-allyl tricarbonyliron lactone complexes serve as key precursor
molecules to install the relative 1,5-carbon to oxygen stereocentres.

## Experimental

${ }^{1} \mathrm{H}$ NMR spectra were recorded in $\mathrm{CDCl}_{3}$ unless stated otherwise on Bruker DPX-200, Bruker DPX-400 or Bruker DRX600 spectrometers and are reported as follows: chemical shift $\delta(\mathrm{ppm})$, (number of protons, multiplicity, coupling constant $J / \mathrm{Hz}$, assignment). Residual protic solvent $\mathrm{CHCl}_{3}\left(\delta_{\mathrm{H}}=\right.$ 7.26 ppm ) was used as the internal reference. ${ }^{13} \mathrm{C}$ NMR spectra were recorded in $\mathrm{CDCl}_{3}$ unless stated otherwise at $50 \mathrm{MHz}, 100$ MHz or 150 MHz on Bruker DPX-200, Bruker DPX-400 or Bruker DRX-600 spectrometers, respectively, using the central resonance of $\mathrm{CDCl}_{3}\left(\delta_{\mathrm{C}}=77.0 \mathrm{ppm}\right)$ as the internal reference. Infra-red spectra were recorded as thin films between sodium chloride plates, deposited from chloroform solution on PerkinElmer 983G or FTIR 1620 spectrometers. Mass spectra were obtained on a Kratos MS890MS spectrometer using electron ionisation or fast atom bombardment techniques, on a Kratos MS50 spectrometer using the fast ion bombardment technique or on a Bruker BIOAPEX 4.7 T FTICR or a Micromass Q-Tof spectrometer using the electrospray technique at the Department of Chemistry, University of Cambridge. Microanalyses were determined in the microanalytical laboratories at the University of Cambridge. Optical rotations were measured with an Optical Activity AA-1000 polarimeter and $[a]_{\mathrm{D}}$ values are given in $10^{-1} \mathrm{deg} \mathrm{cm}{ }^{2} \mathrm{~g}^{-1}$. Flash column chromatography was carried out using Merck Kieselgel ( $230-400$ mesh) unless otherwise indicated. All reactions were carried out under an argon atmosphere in oven-dried glassware which was cooled under a continuous stream of argon immediately prior to use. Reactions involving the preparation of iron complexes were carried out using degassed benzene. The solvent was degassed by successively evacuating and purging the solvent three times with argon while simultaneously subjecting the solvent to sonication using an $80 \mathrm{~W}, 55 \mathrm{kHz}$ cleaning bath. PE refers to petroleum ether, bp $40-60^{\circ} \mathrm{C}$, which was distilled prior to use. $\mathrm{Et}_{2} \mathrm{O}$ and THF were distilled from sodium benzophenone ketyl and DCM; benzene and acetonitrile were distilled from calcium hydride. Other reagents and solvents were used as supplied. Aqueous solutions are saturated unless otherwise specified.

## Diethyl (6-benzyloxy-2-oxo-1-hexyl)phosphonate (6)

A solution of diethyl (2-oxopropyl)phosphonate ( $68.0 \mathrm{~g}, 350$ mmol ) in THF ( 300 ml ) was added dropwise to a stirred slurry of $\mathrm{NaH}[15.3 \mathrm{~g}$ of a $60 \%$ dispersion in mineral oil, prewashed with THF ( $2 \times 30 \mathrm{ml}$ ), 382 mmol$]$ in THF $(375 \mathrm{ml})$ at $0^{\circ} \mathrm{C}$. The reaction mixture was allowed to warm to rt and stirred at this temperature for 1 h . After recooling to $0^{\circ} \mathrm{C},{ }^{\mathrm{n}} \mathrm{BuLi}(153 \mathrm{ml}$ of a 2.5 M solution in hexane, 382 mmol ) was added and the reaction mixture was stirred for another 30 min at $0^{\circ} \mathrm{C}$. A solution of 1-benzyloxy-3-bromopropane ( $72.8 \mathrm{~g}, 318 \mathrm{mmol}$ ) in THF $(300 \mathrm{ml})$ was added dropwise, the reaction mixture was allowed to warm to rt and stirred for 1.5 h . The reaction mixture was carefully poured into ice-cold aqueous $\mathrm{NH}_{4} \mathrm{Cl}$ solution ( 1000 ml ) and the aqueous phase was extracted with $\mathrm{Et}_{2} \mathrm{O}$ $(3 \times 500 \mathrm{ml})$. The combined organic extracts were washed with brine ( 200 ml ), dried over $\mathrm{MgSO}_{4}$ and concentrated under reduced pressure. Purification of the residue by flash column chromatography (silica gel, neat $\mathrm{Et}_{2} \mathrm{O}$ to $5 \% \mathrm{MeOH}$ in $\mathrm{Et}_{2} \mathrm{O}$ ) afforded the phosphonate $6(89 \mathrm{~g}, 82 \%)$ as a light yellow liquid; $v_{\text {max }}($ film $) / \mathrm{cm}^{-1}$ 2981, 2940, 2860, 1713 (C=O), 1453, 1393, 1366, $1251(\mathrm{P}=\mathrm{O}), 1100,1023,965$; $\delta_{\mathrm{H}}(400 \mathrm{MHz}) 7.33-$ $7.20(5 \mathrm{H}, \mathrm{m}, 5 \times \mathrm{Ph}-H), 4.45(2 \mathrm{H}, \mathrm{s}, \mathrm{PhCH}$ ) , 4.15-4.03 ( $4 \mathrm{H}, \mathrm{m}$, $\left.2 \times \mathrm{OCH}_{2} \mathrm{CH}_{3}\right), 3.44(2 \mathrm{H}, \mathrm{t}, J 6.1,2 \times 6-\mathrm{H}), 3.02(2 \mathrm{H}, \mathrm{d}, J 22.8$, $2 \times 1-\mathrm{H}), 2.61(2 \mathrm{H}, \mathrm{t}, J 7.0,2 \times 3-\mathrm{H}), 1.71-1.54(4 \mathrm{H}, \mathrm{m}, 2 \times$ $4-\mathrm{H}, 2 \times 5-\mathrm{H}), 1.29\left(6 \mathrm{H}, \mathrm{t}, J 7.0,2 \times \mathrm{OCH}_{2} \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}(100 \mathrm{MHz})$ 201.8 (d, $\left.{ }^{2} J_{\mathrm{CP}} 6.2,2-\mathrm{C}\right), 138.5$ (quat. Ph-C), 128.3 (Ph-C), 127.6

ii


Scheme 2 Reagents and conditions: i. $(\mathrm{COCl})_{2}, \mathrm{DMSO}, \mathrm{Et}_{3} \mathrm{~N}, \mathrm{DCM},-78{ }^{\circ} \mathrm{C}, 3 \mathrm{~h}, 86 \% \mathbf{1 3 a}, 80 \% \mathbf{1 1 b}$; ii. $\mathrm{Bu}_{3} \mathrm{SnCHI}_{2}, \mathrm{CrCl}_{2}, \mathrm{DMF}, 0{ }^{\circ} \mathrm{C}, 67 \% \mathbf{1 4 a}$, $69 \%$ 14b; iii. methyl 4,6-bis[(2-methoxyethoxy)methyloxy]-2-iodobenzoate 15, $\mathrm{Pd}_{2}(\mathrm{dba})_{3}, \mathrm{P}(2 \text {-furyl })_{3}$, toluene, $100{ }^{\circ} \mathrm{C}, 4 \mathrm{~h}, 82 \%$ 16a, $85 \% \mathbf{1 6 b}$; iv. HFpyridine, pyridine, THF, $12 \mathrm{~h}, 95 \% \mathbf{1 7 a}, 93 \% \mathbf{1 7 b}$; v. 10 M aqueous KOH , ethane-1,2-diol, $120{ }^{\circ} \mathrm{C}, 4 \mathrm{~h}, 87 \% \mathbf{1 8 a}, 91 \% \mathbf{1 8 b}$; vi. syringe pump addition of a solution of the seco acid and $\mathrm{Et}_{3} \mathrm{~N}$ in MeCN over 10 h to 1 -methyl-2-chloropyridinium iodide, MeCN, reflux, $64 \% 19 \mathrm{a}, 62 \% \mathbf{1 9 b}$; vii. 1.5 M aqueous $\mathrm{HCl}, \mathrm{THF}, 40^{\circ} \mathrm{C}, 93 \% \mathbf{1}, 93 \% 2$.
(Ph-C), 127.5 ( $\mathrm{Ph}-\mathrm{C}$ ), $72.9\left(\mathrm{PhCH}_{2}\right), 69.9$ (6-C), 62.5 (d, ${ }^{2} J_{\mathrm{CP}} 6.4,2 \times \mathrm{OCH}_{2} \mathrm{CH}_{3}$ ), 43.7 (d, $\left.{ }^{3} J_{\mathrm{CP}} 0.8,3-\mathrm{C}\right), 42.3$ (d, ${ }^{1} J_{\mathrm{CP}} 126.4,1-\mathrm{C}$ ), 28.9 (4-C or $5-\mathrm{C}$ ), 20.2 ( $5-\mathrm{C}$ or $4-\mathrm{C}$ ), 16.2 (d, $\left.{ }^{3} J_{\mathrm{CP}} 6.2,2 \times \mathrm{OCH}_{2} \mathrm{CH}_{3}\right) ; m / z(\mathrm{EI}) 342\left(\mathrm{M}^{+}, 1 \%\right), 284$ (6), 251 ( $\mathrm{M}^{+}-\mathrm{Bn}, 6$ ), 207 (25), 194 (57), 179 (26), 91 (100) [Found ( $\mathrm{M}^{+}$) 342.1604. $\mathrm{C}_{17} \mathrm{H}_{27} \mathrm{O}_{5} \mathrm{P}$ requires $\mathrm{M}, 342.1596$ ].

## ( $1^{\prime}$ E,4R,5S)-4-(7'-Benzyloxy-3'-oxohept-1'-en-1'-yl)-2,2,5-trimethyl-1,3-dioxolane (7)

$\mathrm{LiAlH}_{4}\left(69.0 \mathrm{ml}\right.$ of a 1 M solution in $\left.\mathrm{Et}_{2} \mathrm{O}, 69 \mathrm{mmol}\right)$ was added dropwise to a solution of the ester $5(20.0 \mathrm{~g}, 115 \mathrm{mmol})$ in $\mathrm{Et}_{2} \mathrm{O}$ $(120 \mathrm{ml})$ at $0{ }^{\circ} \mathrm{C}$ and the reaction mixture was stirred at this temperature for 2 h . Water ( 2.6 ml ) was added followed by $15 \%$ NaOH solution ( 2.6 ml ) and water ( 7.8 ml ) and the reaction mixture was stirred at rt for 12 h . The reaction mixture was filtered, washing the residue with $\mathrm{Et}_{2} \mathrm{O}(2 \times 50 \mathrm{ml})$. Concentration of the filtrate under reduced pressure afforded the crude
alcohol which was used without further purification. A solution of DMSO ( $21.2 \mathrm{ml}, 299 \mathrm{mmol}$ ) in DCM ( 280 ml ) was added dropwise to a solution of $(\mathrm{COCl})_{2}(13.0 \mathrm{ml}, 150 \mathrm{mmol})$ in DCM $(280 \mathrm{ml})$ at $-78^{\circ} \mathrm{C}$ and the reaction mixture was stirred for 30 min . A solution of the crude alcohol in DCM ( 280 ml ) was added dropwise and after $30 \mathrm{~min} \mathrm{Et}_{3} \mathrm{~N}(56.0 \mathrm{ml}, 403 \mathrm{mmol})$ was added dropwise at $-78^{\circ} \mathrm{C}$. Stirring was continued at $-78^{\circ} \mathrm{C}$ for 1 h before the mixture was allowed to warm to rt . The reaction mixture was concentrated under reduced pressure, taken up in $\mathrm{Et}_{2} \mathrm{O}(200 \mathrm{ml})$ and filtered through a plug of silica gel, washing the residue with $\mathrm{Et}_{2} \mathrm{O}(500 \mathrm{ml})$. Concentration of the filtrate under reduced pressure afforded the corresponding crude aldehyde which was used without further purification. A solution of the phosphonate $6(79.0 \mathrm{~g}, 230 \mathrm{mmol})$ in THF ( 140 ml ) was added dropwise to a stirred slurry of $\mathrm{NaH}[9.2 \mathrm{~g}$ of a $60 \%$ dispersion in mineral oil, prewashed with THF $(2 \times 5 \mathrm{ml})$, $230 \mathrm{mmol}]$ in THF $(140 \mathrm{ml})$ at $0^{\circ} \mathrm{C}$. The reaction mixture was allowed to warm to rt and stirred at this temperature for 1 h .

After recooling to $-78^{\circ} \mathrm{C}$ a solution of the crude aldehyde in THF ( 140 ml ) was added and the reaction mixture was stirred for 1 h . The reaction mixture was quenched by addition of $\mathrm{NH}_{4} \mathrm{Cl}$ solution ( 20 ml ), allowed to warm to rt and poured into aqueous $\mathrm{NH}_{4} \mathrm{Cl}$ solution ( 400 ml ). The aqueous phase was extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 300 \mathrm{ml})$. The organic extracts were washed with brine ( 150 ml ), dried over $\mathrm{MgSO}_{4}$ and concentrated under reduced pressure. Purification of the residue by flash column chromatography (silica gel, 10 to $50 \% \mathrm{Et}_{2} \mathrm{O}$ in PE) afforded the enone $7(31.8 \mathrm{~g}, 83 \%$ over three steps) as a light yellow liquid; $[a]_{\mathrm{D}}^{31}+6.6$ ( c 1.00 in $\mathrm{CHCl}_{3}$ ) (Found: C, 72.4; H , 8.6. $\mathrm{C}_{20} \mathrm{H}_{28} \mathrm{O}_{4}$ requires C, $72.3 ; \mathrm{H}, 8.5 \%$ ); $v_{\text {max }}($ film $) / \mathrm{cm}^{-1} 2984$, 2933, 2866, 1698, 1678 (C=O), 1638 (C=C), 1454, 1379, 1240 , 1173, 1105, 1036, 979, 858, 736, 698; $\delta_{\mathrm{H}}(600 \mathrm{MHz}) 7.34-7.22$ $(5 \mathrm{H}, \mathrm{m}, 5 \times \mathrm{Ph}-H), 6.67\left(1 \mathrm{H}, \mathrm{dd}, J 15.9,5.8,1^{\prime}-\mathrm{H}\right), 6.35(1 \mathrm{H}$, dd, $J 15.9,1.0,2-\mathrm{H}), 4.46\left(2 \mathrm{H}, \mathrm{s}, \mathrm{PhCH}_{2}\right), 4.04(1 \mathrm{H}, \mathrm{ddd}, J 8.2$, $5.8,1.0,4-\mathrm{H}), 3.83-3.77(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}), 3.46(2 \mathrm{H}, \mathrm{t}, J 6.2,2 \times$ $\left.7^{\prime}-\mathrm{H}\right), 2.57\left(2 \mathrm{H}, \mathrm{t}, J 7.3,2 \times 4^{\prime}-\mathrm{H}\right), 1.74-1.67\left(2 \mathrm{H}, \mathrm{m}, 2 \times 5^{\prime}-\mathrm{H}\right)$, $1.66-1.58\left(2 \mathrm{H}, \mathrm{m}, 2 \times 6^{\prime}-\mathrm{H}\right), 1.43\left(3 \mathrm{H}, \mathrm{s}, 2-\mathrm{CH}_{3}\right), 1.40(3 \mathrm{H}, \mathrm{s}$, $\left.2-\mathrm{CH}_{3}\right), 1.28\left(3 \mathrm{H}, \mathrm{d}, J 6.1,5-\mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}(150 \mathrm{MHz}) 199.6\left(3^{\prime}-\mathrm{C}\right)$, 140.9 (1'-C), 138.5 (quat. Ph-C), 130.5 ( $2^{\prime}-\mathrm{C}$ ), 128.3 (Ph-C), 127.6 (Ph-C), 127.5 (Ph-C), 109.2 (2-C), 81.9 (4-C), 76.5 (5-C), $72.8\left(\mathrm{PhCH}_{2}\right), 69.9$ ( $7^{\prime}-\mathrm{C}$ ), 40.4 ( $\left.4^{\prime}-\mathrm{C}\right), 29.1$ ( $\left.6^{\prime}-\mathrm{C}\right), 27.3$ $\left(2-\mathrm{CH}_{3}\right), 26.7\left(2-\mathrm{CH}_{3}\right), 20.7\left(5^{\prime}-\mathrm{C}\right), 16.7\left(5-\mathrm{CH}_{3}\right) ; m / z$ (EI) 332 $\left(\mathrm{M}^{+}, 1 \%\right), 214$ (41), 185 (46), 172 (53), 169 (45), 127 (54), 111 (66), 97 (100), 91 (43) [Found ( $\mathrm{M}^{+}$) 332.1997. $\mathrm{C}_{20} \mathrm{H}_{28} \mathrm{O}_{4}$ requires M, 332.1988].

## ( $6 E, \mathbf{8 S}, 9 \mathrm{~S}$ )-1-Benzyloxy-8,9-dihydroxydec-6-en-5-one (8)

$\mathrm{AcOH}(75 \mathrm{ml})$ and water $(75 \mathrm{ml})$ were added to the acetonide 7 $(15.0 \mathrm{~g}, 45 \mathrm{mmol})$ and the reaction mixture was stirred at $40^{\circ} \mathrm{C}$ for 20 h . The reaction mixture was carefully poured into an aqueous $\mathrm{NaHCO}_{3}$ solution ( 300 ml ) and solid $\mathrm{NaHCO}_{3}$ was added until effervescence stopped. The mixture was extracted with EtOAc ( $4 \times 300 \mathrm{ml}$ ), the combined organic extracts were washed with brine ( 100 ml ), dried over $\mathrm{MgSO}_{4}$ and concentrated under reduced pressure. Purification of the residue by flash column chromatography ( $75 \%$ EtOAc in PE to neat EtOAc) afforded the diol $8(12.1 \mathrm{~g}, 92 \%)$ as a colourless oil; $[a]_{D}^{30}$ $-10.6\left(c 1.00\right.$ in $\left.\mathrm{CHCl}_{3}\right) ; v_{\text {max }}($ film $) / \mathrm{cm}^{-1} 3420(\mathrm{OH}), 2936,2867$, 1654 ( $\mathrm{C}=\mathrm{O}$ ), 1638 ( $\mathrm{C}=\mathrm{C}$ ), 1558, 1540, 1507, 1496, 1456, 1365, $1319,1154,1076,986 ; \delta_{\mathrm{H}}(400 \mathrm{MHz}) 7.37-7.24(5 \mathrm{H}, \mathrm{m}, 5 \times$ $\mathrm{Ph}-H), 6.75$ ( $1 \mathrm{H}, \mathrm{dd}, J 15.9,5.1,7-\mathrm{H}), 6.38$ ( 1 H , dd, J 15.9, 1.4, $6-\mathrm{H}), 4.48\left(2 \mathrm{H}, \mathrm{s}, \mathrm{PhCH}_{2}\right), 4.06-3.98(1 \mathrm{H}, \mathrm{m}, 8-\mathrm{H}), 3.73-3.64$ $(1 \mathrm{H}, \mathrm{m}, 9-\mathrm{H}), 3.48(2 \mathrm{H}, \mathrm{t}, J 6.3,2 \times 1-\mathrm{H}), 2.99(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH})$, $2.66(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 2.58(2 \mathrm{H}, \mathrm{t}, J 7.3,2 \times 4-\mathrm{H}), 1.76-1.58(4 \mathrm{H}$, $\mathrm{m}, 2 \times 2-\mathrm{H}, 2 \times 3-\mathrm{H}), 1.21(3 \mathrm{H}, \mathrm{d}, J 6.3,3 \times 10-\mathrm{H}) ; \delta_{\mathrm{C}}(100$ $\mathrm{MHz}) 200.3$ (5-C), 144.2 (7-C), 138.4 (quat. Ph-C), 130.1 (6-C), 128.4 ( $\mathrm{Ph}-\mathrm{C}$ ), 127.7 ( $\mathrm{Ph}-\mathrm{C}$ ), 127.6 ( $\mathrm{Ph}-\mathrm{C}$ ), 75.7 ( $8-\mathrm{C}$ ), 72.9 $\left(\mathrm{PhCH}_{2}\right), 70.3$ (9-C), 70.0 (1-C), 40.6 (4-C), 29.1 (2-C or 3-C), 20.8 (3-C or $2-\mathrm{C}$ ), 19.2 ( $10-\mathrm{C}$ ); $m / z$ (EI) 293 ( $\mathrm{MH}^{+}, 11 \%$ ), 185 (8), 168 (59), 91 (100) [Found $\left(\mathrm{MH}^{+}\right)$293.1753. $\mathrm{C}_{17} \mathrm{H}_{25} \mathrm{O}_{4}$ requires MH, 293.1753]

## (1'E,2RS,4S,5S)-4-(7'-Benzyloxy-3'-oxohept-1'-en-1'-yl)-5-methyl-1,3-dioxa-2-thiolane 2-oxide (9)

Freshly distilled $\mathrm{SOCl}_{2}(2.11 \mathrm{ml}, 26.9 \mathrm{mmol})$ was added dropwise to a solution of the $\operatorname{diol} \mathbf{8}(5.23 \mathrm{~g}, 17.9 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}$ $(7.47 \mathrm{ml}, 53.7 \mathrm{mmol})$ in $\mathrm{Et}_{2} \mathrm{O}(100 \mathrm{ml})$ at $0^{\circ} \mathrm{C}$. After 90 min the reaction mixture was poured into aqueous $\mathrm{NaHCO}_{3}$ solution $(100 \mathrm{ml})$, the phases were separated and the aqueous phase was extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 100 \mathrm{ml})$. The combined organic extracts were washed with brine ( 40 ml ), dried over $\mathrm{MgSO}_{4}$ and concentrated under reduced pressure. Purification of the residue by flash column chromatography (silica gel, $50 \% \mathrm{Et}_{2} \mathrm{O}$ in PE) afforded the cyclic sulfite $\mathbf{9}(5.37 \mathrm{~g}, 89 \%)$ as a yellow oil; (1:1 mixture of diastereoisomers at sulfur); $v_{\max }$ (film) $/ \mathrm{cm}^{-1}$ 2933, 2847, 1677 (C=O), 1640 (C=C), 1451, 1358, 1209, 1100, 1046, 956, 867; $\delta_{\mathrm{H}}(200 \mathrm{MHz}) 7.40-7.22(5 \mathrm{H}, \mathrm{m}, 5 \times \mathrm{Ph}-H)$,
$\{[6.75$ (dd, $J 15.8,6.3), 6.64$ (dd, $\left.J 15.8,5.9)] 1 \mathrm{H}, 1^{\prime}-\mathrm{H}\right\},\{[6.45$ (dd, $J 15.8,1.0$ ), 6.43 (dd, $\left.J 15.8,1.0)] 1 \mathrm{H}, 2^{\prime}-\mathrm{H}\right\},\{[5.05$ (ddd, $J 8.8,5.9,1.0), 4.55-4.44(\mathrm{~m})] 1 \mathrm{H}, 4-\mathrm{H}\},\{[4.79(\mathrm{dq}, J 9.1,6.1)$, 4.27 (dq, J 8.8, 6.3)] 1H, 5-H\}, 4.49 ( $2 \mathrm{H}, \mathrm{s}, \mathrm{PhCH}_{2}$ ), 3.55-3.43 $\left(2 \mathrm{H}, \mathrm{m}, 2 \times 7^{\prime}-\mathrm{H}\right), 2.69-2.54\left(2 \mathrm{H}, \mathrm{m}, 2 \times 4^{\prime}-\mathrm{H}\right), 1.85-1.55(4 \mathrm{H}$, $\left.\mathrm{m}, 2 \times 5^{\prime}-\mathrm{H}, 2 \times 6^{\prime}-\mathrm{H}\right),\{[1.59(\mathrm{~d}, J 6.3), 1.51(\mathrm{~d}, J 6.1)] 3 \mathrm{H}$, $\left.5-\mathrm{CH}_{3}\right\} ; \delta_{\mathrm{C}}(50 \mathrm{MHz})$ [198.8, 198.6 (3'-C)], 138.4 (quat. Ph-C), $\left[136.8,134.8,132.9,132.6\right.$ ( $\left.\left.1^{\prime}-\mathrm{C}, 2^{\prime}-\mathrm{C}\right)\right], 128.3$ (Ph-C), 127.6 (Ph-C), 127.5 (Ph-C), [87.0, 83.7, 82.6, 79.0 (4-C, 5-C)], 72.9 $\left(\mathrm{PhCH}_{2}\right), 69.8$ (7'-C), [41.1, 40.7 (4'-C)], [29.0, 20.6, 20.5 ( $5^{\prime}-\mathrm{C}$, $\left.\left.6^{\prime}-\mathrm{C}\right)\right],\left[18.0,15.7\left(5-\mathrm{CH}_{3}\right)\right] ; m / z(\mathrm{FAB}) 361\left(\mathrm{MNa}^{+}, 10 \%\right), 339$ $\left(\mathrm{MH}^{+}, 68\right), 231\left(\mathrm{M}^{+}-\mathrm{Bn}, 38\right)$ [Found $\left(\mathrm{MH}^{+}\right)$339.1265. $\mathrm{C}_{17} \mathrm{H}_{23} \mathrm{O}_{5} \mathrm{~S}$ requires $\mathrm{MH}, 339.1266$ ].
[(4E,2S,3R)-2-(Carbonyloxy-кC)-10-benzyloxy-6-oxo-(3,4,5-ף)-dec-4-en-3-yl]tricarbonyliron (10a) and [( $4 E, 2 S, 3 S$ )-2-(carbonyloxy-к $C$ )-10-benzyloxy-6-oxo-(3,4,5-ๆ)-dec-4-en-3-yl]tricarbonyliron (10b)
A solution of the cyclic sulfite $\mathbf{9}(5.37 \mathrm{~g}, 15.9 \mathrm{mmol})$ in degassed $\mathrm{PhH}(300 \mathrm{ml})$ was added to $\mathrm{Fe}_{2}(\mathrm{CO})_{9}(17.32 \mathrm{~g}, 47.6 \mathrm{mmol})$ and the reaction mixture was submitted to sonication at rt for 5 h . The mixture was filtered through a plug of Celite, washing the residue with $\mathrm{Et}_{2} \mathrm{O}(1500 \mathrm{ml})$. After concentration of the filtrate under reduced pressure the residue was purified by flash column chromatography (silica gel, 1 to $70 \% \mathrm{Et}_{2} \mathrm{O}$ in PE ) affording a 1:1 mixture of the endo complex 10a and the exo complex 10b $(5.20 \mathrm{~g}, 74 \%)$. Separation of the diastereoisomeric complexes was achieved using a Biotage FLASH 40i system (FLASH 40M cartridge, $65 \% \mathrm{Et}_{2} \mathrm{O}$ in PE ) giving, in order of elution, the endo complex $10 \mathrm{a}(2.46 \mathrm{~g}, 35 \%)$ as a bright yellow oil; $[a]_{\mathrm{D}}^{27}+369.0$ (c 1.00 in DCM); $v_{\max }$ (film)/ $\mathrm{cm}^{-1}$ 2934, 2860, 2088 (CO), 2023 (CO), 1675 (C=O), 1496, 1454, 1373, 1359, 1301, 1180, 1087, 1047, 996, 945, 831, 737, 697; $\delta_{\mathrm{H}}(600 \mathrm{MHz}) 7.38-7.24(5 \mathrm{H}, \mathrm{m}$, $5 \times \mathrm{Ph}-H$ ), 5.53 ( $1 \mathrm{H}, \mathrm{dd}, J 11.2,8.6,4-\mathrm{H}$ ), 5.04 ( $1 \mathrm{H}, \mathrm{dd}, J 8.6$, 4.6, 3-H), $4.56-4.47$ ( $3 \mathrm{H}, \mathrm{m}$ [incl. $4.50,2 \mathrm{H}, \mathrm{s}, \mathrm{PhCH}_{2}$ ] 2-H, $\left.\mathrm{PhCH}_{2}\right), 3.87(1 \mathrm{H}, \mathrm{d}, J 11.2,5-\mathrm{H}), 3.50(2 \mathrm{H}, \mathrm{t}, J 6.3,2 \times 10-\mathrm{H})$, $2.82-2.69(2 \mathrm{H}, \mathrm{m}, 2 \times 7-\mathrm{H}), 1.90-1.65(4 \mathrm{H}, \mathrm{m}, 2 \times 8-\mathrm{H}, 2 \times$ $9-\mathrm{H}), 1.37$ ( $3 \mathrm{H}, \mathrm{d}, J 6.4,3 \times 1-\mathrm{H}$ ); $\delta_{\mathrm{C}}(150 \mathrm{MHz}) 207.9$ (CO), 205.0 (CO), 203.9 (CO), 202.6 (CO), 199.7 (CO), 138.5 (quat. $\mathrm{Ph}-\mathrm{C}$ ), 128.4 ( $\mathrm{Ph}-\mathrm{C}$ ), 127.6 ( $2 \times \mathrm{Ph}-\mathrm{C}$ ), $92.0(\mathrm{CH}), 85.4(\mathrm{CH})$, $73.0(\mathrm{CH}), 72.9\left(\mathrm{PhCH}_{2}\right), 69.8(10-\mathrm{C}), 65.8(\mathrm{CH}), 43.0(7-\mathrm{C})$, 29.1 ( $8-\mathrm{C}$ or $9-\mathrm{C}$ ), 21.8 (1-C), 20.6 ( $9-\mathrm{C}$ or $8-\mathrm{C}$ ); $m / z$ (FIB) $443\left(\mathrm{MH}^{+}, 75 \%\right), 386\left(\mathrm{M}^{+}-2 \mathrm{CO}, 35\right), 358\left(\mathrm{M}^{+}-3 \mathrm{CO}, 32\right)$, $331\left(\mathrm{MH}^{+}-4 \mathrm{CO}, 100\right), 239\left(\mathrm{M}^{+}-4 \mathrm{CO}-\mathrm{Bn}, 13\right), 223$ (45) [Found ( $\mathrm{MH}^{+}$) 443.0785. $\mathrm{C}_{21} \mathrm{H}_{23} \mathrm{FeO}_{7}$ requires $\mathrm{MH}, 443.0793$ ]; and the exo complex $\mathbf{1 0 b}(2.46 \mathrm{~g}, 35 \%)$ as a bright yellow oil; $[a]_{\mathrm{D}}^{27}-339.5$ ( $c 1.00$ in DCM); $v_{\max }$ (film) $/ \mathrm{cm}^{-1} 3033,2933$, 2861, 2082 (CO), 2017 (CO), 1665 (C=O), 1496, 1453, 1305, 1120, 1104, 1049, 995, 946, 739, 695, 652, 603; $\delta_{\mathrm{H}}(600 \mathrm{MHz}) 7.37-$ $7.24(5 \mathrm{H}, \mathrm{m}, 5 \times \mathrm{Ph}-H), 5.68$ ( $1 \mathrm{H}, \mathrm{dd}, J 11.0,8.4,4-\mathrm{H}$ ), 4.80 $(1 \mathrm{H}, \mathrm{d}, J 8.4,3-\mathrm{H}), 4.50\left(2 \mathrm{H}, \mathrm{s}, \mathrm{PhCH}_{2}\right), 4.27(1 \mathrm{H}, \mathrm{q}, J 6.5$, $2-\mathrm{H}), 3.72(1 \mathrm{H}, \mathrm{d}, J 11.0,5-\mathrm{H}), 3.50(2 \mathrm{H}, \mathrm{t}, J 6.4,2 \times 10-\mathrm{H})$, $2.79-2.66(2 \mathrm{H}, \mathrm{m}, 2 \times 7-\mathrm{H}), 1.85-1.63(4 \mathrm{H}, \mathrm{m}, 2 \times 8-\mathrm{H}, 2 \times$ $9-\mathrm{H}), 1.42(3 \mathrm{H}, \mathrm{d}, J 6.5,3 \times 1-\mathrm{H}) ; \delta_{\mathrm{C}}(150 \mathrm{MHz}) 208.1(\mathrm{CO})$, 204.9 (CO), 203.9 (CO), 202.5 (CO), 200.0 (CO), 138.5 (quat. Ph-C), 128.3 (Ph-C), 127.6 (Ph-C), 127.5 (Ph-C), 93.6 (CH), $84.4(\mathrm{CH}), 72.9\left(\mathrm{PhCH}_{2}\right), 70.7(\mathrm{CH}), 69.8(10-\mathrm{C}), 65.1(\mathrm{CH})$, $42.9\left(\mathrm{CH}_{2}\right), 29.1$ ( $8-\mathrm{C}$ or $9-\mathrm{C}$ ), $24.0(1-\mathrm{C}), 20.6$ ( $9-\mathrm{C}$ or $8-\mathrm{C}$ ); $m / z$ (FIB) $443\left(\mathrm{MH}^{+}, 28 \%\right), 386\left(\mathrm{M}^{+}-2 \mathrm{CO}, 13\right), 359\left(\mathrm{MH}^{+}-\right.$ $3 \mathrm{CO}, 7), 331\left(\mathrm{MH}^{+}-4 \mathrm{CO}, 100\right), 239\left(\mathrm{M}^{+}-4 \mathrm{CO}-\mathrm{Bn}, 10\right)$, 223 (44) [Found ( $\mathrm{MH}^{+}$) 443.0811. $\mathrm{C}_{21} \mathrm{H}_{23} \mathrm{FeO}_{7}$ requires MH, 443.0793].

## General procedure for the preparation of alcohol complexes 11a and 11b

For a 1.45 mmol scale reaction: $\mathrm{Al}^{\mathrm{n}} \mathrm{Pr}_{3}(1.0 \mathrm{M}$ solution in $\mathrm{PhMe}, 3.0$ equiv.) was added dropwise to a solution of the ketone complex ( 1.0 equiv.) in DCM ( 20 ml ) at $0^{\circ} \mathrm{C}$. After 1 h 1 M aqueous HCl solution $(10 \mathrm{ml})$ was added and the mixture
was partioned between water ( 30 ml ) and $\mathrm{Et}_{2} \mathrm{O}(30 \mathrm{ml})$. The aqueous phase was extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 30 \mathrm{ml})$, the combined organic extracts were washed with brine ( 30 ml ), dried over $\mathrm{MgSO}_{4}$ and concentrated under reduced pressure. Purification of the residue by flash column chromatography (silica gel, 50 to $75 \% \mathrm{Et}_{2} \mathrm{O}$ in PE) afforded the alcohol complex.

## [(4E,2S,3R,6R)-2-(Carbonyloxy-кC)-10-benzyloxy-6-hydroxy-(3,4,5- $\boldsymbol{\eta}$ )-dec-4-en-3-yl]tricarbonyliron (11a)

Prepared according to the general procedure using endo complex 10a ( $628 \mathrm{mg}, 1.42 \mathrm{mmol}$ ). The alcohol complex 11a ( 595 $\mathrm{mg}, 94 \%$ ) was obtained as a yellow oil; $[a]_{\mathrm{D}}^{30}+115.0$ (c 1.00 in $\mathrm{CHCl}_{3}$ ); $v_{\text {max }}($ film $) / \mathrm{cm}^{-1} 3417(\mathrm{OH}), 2933,2862,2079(\mathrm{CO})$, 2023 (CO), 2002 (CO), 1665 (C=O), 1638, 1450, 1360, 1082, $1039,995,946,831,739,695,657 ; \delta_{\mathrm{H}}(600 \mathrm{MHz}) 7.38-7.24(5 \mathrm{H}$, $\mathrm{m}, 5 \times \mathrm{Ph}-H), 4.75(1 \mathrm{H}, \mathrm{dd}, J 12.2,8.3,4-\mathrm{H}), 4.60(1 \mathrm{H}, \mathrm{dd}$, $J 8.3,4.7,3-\mathrm{H}), 4.50\left(2 \mathrm{H}, \mathrm{s}, \mathrm{PhCH}_{2}\right), 4.46-4.40(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H})$, $4.14-4.08(1 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}), 4.03(1 \mathrm{H}, \mathrm{dd}, J 12.2,3.6,5-\mathrm{H}), 3.52$ $(2 \mathrm{H}, \mathrm{t}, J 6.1,2 \times 10-\mathrm{H}), 2.54-2.45(1 \mathrm{H}, \mathrm{m}, \mathrm{OH}), 1.84-1.50(6 \mathrm{H}$, $\mathrm{m}, 2 \times 7-\mathrm{H}, 2 \times 8-\mathrm{H}, 2 \times 9-\mathrm{H}), 1.34(3 \mathrm{H}, \mathrm{d}, J 6.4,3 \times 1-\mathrm{H})$; $\delta_{\mathrm{C}}(150 \mathrm{MHz}) 209.5(\mathrm{CO}), 206.6(2 \times \mathrm{CO}), 203.4(\mathrm{CO}), 138.3$ (quat. $\mathrm{Ph}-\mathrm{C}$ ), 128.4 ( $\mathrm{Ph}-\mathrm{C}$ ), 127.7 ( $\mathrm{Ph}-\mathrm{C}$ ), 127.6 ( $\mathrm{Ph}-\mathrm{C}$ ), 88.5 $(\mathrm{CH}), 88.0(\mathrm{CH}), 76.8(\mathrm{CH}), 73.3(\mathrm{CH}), 73.0\left(\mathrm{PhCH}_{2}\right), 71.7$ (CH), 70.1 ( $10-\mathrm{C}$ ), 39.4 ( $7-\mathrm{C}$ ), 29.2 ( $8-\mathrm{C}$ or 9-C), 22.8 ( $9-\mathrm{C}$ or 8 -C), 21.8 (1-C); $m / z$ (FIB) $445\left(\mathrm{MH}^{+}, 12 \%\right), 387$ (11), 371 $\left(\mathrm{MH}^{+}-2 \mathrm{CO}-\mathrm{H}_{2} \mathrm{O}, 7\right), 333\left(\mathrm{MH}^{+}-4 \mathrm{CO}, 13\right), 315\left(\mathrm{MH}^{+}-\right.$ $4 \mathrm{CO}-\mathrm{H}_{2} \mathrm{O}, 100$ ) [Found $\left(\mathrm{MH}^{+}\right)$445.0976. $\mathrm{C}_{21} \mathrm{H}_{25} \mathrm{FeO}_{7}$ requires MH, 445.0950].

## [( $4 E, 2 S, 3 S, 6 S$ )-2-(Carbonyloxy-кC)-10-benzyloxy-6-hydroxy-(3,4,5-ף)-dec-4-en-3-yl]tricarbonyliron (11b)

Prepared according to the general procedure using exo complex 10b ( $655 \mathrm{mg}, 1.48 \mathrm{mmol}$ ). The alcohol complex 11b ( 524 mg , $80 \%$ ) was obtained as a yellow oil; $[a]_{\mathrm{D}}^{30}+77.0\left(c 1.00\right.$ in $\left.\mathrm{CHCl}_{3}\right)$; $v_{\max }($ film $) / \mathrm{cm}^{-1} 3418(\mathrm{OH}), 2936,2862,2076(\mathrm{CO}), 2025(\mathrm{CO})$, 2003 (CO), 1660 (C=O), 1634, 1450, 1377, 1333, 1307, 1090, 1046, 1007, 950, 736, 698; $\delta_{\mathrm{H}}(600 \mathrm{MHz}) 7.37-7.25(5 \mathrm{H}, \mathrm{m}, 5 \times$ $\mathrm{Ph}-H), 4.90(1 \mathrm{H}, \mathrm{dd}, J 12.1,8.0,4-\mathrm{H}), 4.49\left(2 \mathrm{H}, \mathrm{s}, \mathrm{PhCH}_{2}\right)$, $4.37(1 \mathrm{H}, \mathrm{d}, J 8.0,3-\mathrm{H}), 4.21(1 \mathrm{H}, \mathrm{q}, J 6.4,2-\mathrm{H}), 4.11-4.02(1 \mathrm{H}$, $\mathrm{m}, 6-\mathrm{H}), 3.89(1 \mathrm{H}, \mathrm{dd}, J 12.1,3.8,5-\mathrm{H}), 3.51(2 \mathrm{H}, \mathrm{t}, J 6.2,2 \times$ $10-\mathrm{H}), 2.77-2.69(1 \mathrm{H}, \mathrm{m}, \mathrm{OH}), 1.78-1.48(6 \mathrm{H}, \mathrm{m}, 2 \times 7-\mathrm{H}, 2 \times$ $8-\mathrm{H}, 2 \times 9-\mathrm{H}), 1.35(3 \mathrm{H}, \mathrm{d}, J 6.4,3 \times 1-\mathrm{H}) ; \delta_{\mathrm{C}}(150 \mathrm{MHz}) 209.8$ (CO), 206.5 (CO), 206.1 (CO), 203.6 (CO), 138.3 (quat. Ph-C), 128.4 (Ph-C), 127.7 (Ph-C), 127.6 (Ph-C), 89.5 (4-C), 87.6 (5-C), 76.2 (3-C), $73.0\left(\mathrm{PhCH}_{2}\right)$, 71.5 (6-C), 71.2 (2-C), 70.1 (10-C), 39.2 (7-C), 29.2 ( $8-\mathrm{C}$ or $9-\mathrm{C}$ ), 23.8 (1-C), 22.8 ( 9 -C or $8-\mathrm{C}$ ); $m / z$ (FIB) $445\left(\mathrm{MH}^{+}, 6 \%\right), 371\left(\mathrm{MH}^{+}-2 \mathrm{CO}-\mathrm{H}_{2} \mathrm{O}, 10\right), 315$ $\left(\mathrm{MH}^{+}-4 \mathrm{CO}-\mathrm{H}_{2} \mathrm{O}, 91\right), 133$ (100) [Found $\left(\mathrm{MH}^{+}\right) 445.0936$. $\mathrm{C}_{21} \mathrm{H}_{25} \mathrm{FeO}_{7}$ requires MH, 445.0950].

## General procedure for the preparation of alcohols 12a and 12b

For a 1.2 mmol scale reaction: $\mathrm{NaBH}(\mathrm{OAc})_{3}$ ( 5.0 equiv.) was added to a solution of the iron complex (1.0 equiv.) in THF $(10 \mathrm{ml})$ at rt . After three days, acetone $(10 \mathrm{ml})$ was added and the mixture was stirred for 1 h . The suspension was filtered through a plug of Celite, washing the residue with acetone $(50 \mathrm{ml})$ and the filtrate was concentrated under reduced pressure. Purification of the residue by flash column chromatography (silica gel, $65 \%$ EtOAc in PE) afforded the diol (mixture of at least three double bond isomers).

TBDMSCl (4.0 equiv.) was added to a solution of the diol ( 1.0 equiv.) and imidazole ( 5.0 equiv.) in DMF ( 2 ml ) at $0^{\circ} \mathrm{C}$. The mixture was stirred at $0^{\circ} \mathrm{C}$ for 30 min and then at rt for 24 h . Water ( 15 ml ) was added and the mixture was extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 15 \mathrm{ml})$. The combined organic extracts were washed with brine ( 10 ml ), dried over $\mathrm{MgSO}_{4}$ and concentrated under reduced pressure. Purification of the residue by flash
column chromatography (silica gel, $5 \% \mathrm{Et}_{2} \mathrm{O}$ in PE) afforded the bis-silyl ether (mixture of at least three double bond isomers).

Palladium on activated carbon ( 0.1 equiv., $10 \% \mathrm{Pd}$ ) was added to a solution of the benzyl ether ( 1.0 equiv.) in EtOAc $(5 \mathrm{ml})$ and stirred under a hydrogen atmosphere for 12 h . The reaction mixture was filtered through Celite, washing the residue with EtOAc ( 30 ml ) and the filtrate was concentrated under reduced pressure. Purification of the residue by flash column chromatography (silica gel, $30 \% \mathrm{Et}_{2} \mathrm{O}$ in PE) afforded the alcohol.
(5R,9S)-5,9-Bis(tert-butyldimethylsilyloxy)decan-1-ol (12a). Prepared according to the general procedure using endo complex $11 \mathrm{a}(0.52 \mathrm{~g}, 1.2 \mathrm{mmol})$. The alcohol $\mathbf{1 2 a}(268 \mathrm{mg}, 61 \%$ over three steps) was obtained as a colourless oil; $[a]_{\mathrm{D}}^{31}+11.6$ (c 1.00 in $\mathrm{CHCl}_{3}$ ) (Found: C, 63.3; H, 11.8. $\mathrm{C}_{22} \mathrm{H}_{50} \mathrm{O}_{3} \mathrm{Si}_{2}$ requires C, 63.1; H, 12.0\%); $v_{\max }($ film $) / \mathrm{cm}^{-1} 3358(\mathrm{OH}), 2953,2930,2857$, 1472, 1462, 1406, 1375, 1360, 1254, 1129, 1110, 1052, 1005, 938, 835,$773 ; \delta_{\mathrm{H}}(200 \mathrm{MHz}) 3.87-3.54$ ( $4 \mathrm{H}, \mathrm{m}$ [incl. 3.61, $2 \mathrm{H}, \mathrm{t}, J 6.4$, $2 \times 1-\mathrm{H}] 2 \times 1-\mathrm{H}, 5-\mathrm{H}, 9-\mathrm{H}), 1.71(1 \mathrm{H}, \mathrm{br}$ s, OH$), 1.64-1.18$ $(12 \mathrm{H}, \mathrm{m}, 2 \times 2-\mathrm{H}, 2 \times 3-\mathrm{H}, 2 \times 4-\mathrm{H}, 2 \times 6-\mathrm{H}, 2 \times 7-\mathrm{H}, 2 \times 8-\mathrm{H})$, $1.10(3 \mathrm{H}, \mathrm{d}, J 6.1,3 \times 10-\mathrm{H}), 0.87\left(18 \mathrm{H}, \mathrm{s}, 2 \times \operatorname{SiC}\left(\mathrm{CH}_{3}\right)_{3}\right), 0.03$ $\left(6 \mathrm{H}, 2, \mathrm{Si}\left(\mathrm{CH}_{3}\right)_{2}\right), 0.02\left(6 \mathrm{H}, 2, \mathrm{Si}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta_{\mathrm{C}}(50 \mathrm{MHz}) 72.1$ $(\mathrm{CH}), 68.6(\mathrm{CH}), 62.8(1-\mathrm{C}), 39.9\left(\mathrm{CH}_{2}\right), 37.1\left(\mathrm{CH}_{2}\right), 36.7$ $\left(\mathrm{CH}_{2}\right), 32.9\left(\mathrm{CH}_{2}\right), 25.8\left(2 \times \mathrm{SiC}\left(\mathrm{CH}_{3}\right)_{3}\right), 23.7(10-\mathrm{C}), 21.4$ $\left.\left(\mathrm{CH}_{2}\right), 21.3\left(\mathrm{CH}_{2}\right), 18.0\left(2 \times \mathrm{SiC}\left(\mathrm{CH}_{3}\right)_{3}\right),-4.5(2 \times \mathrm{SiCH})_{3}\right)$, $-4.8\left(2 \times \mathrm{SiCH}_{3}\right) ; m / z(\mathrm{FIB}) 419\left(\mathrm{MH}^{+}, 7 \%\right), 345$ (10), 287 (19), 229 (33), 185 (31), 159 (61), 155 (100), 137 (65), 115 (64) [Found $\left(\mathrm{MH}^{+}\right)$419.3410. $\mathrm{C}_{22} \mathrm{H}_{51} \mathrm{O}_{3} \mathrm{Si}_{2}$ requires $\mathrm{MH}, 419.3377$ ].
(5S,9S)-5,9-Bis(tert-butyldimethylsilyloxy)decan-1-ol (12b). Prepared according to the general procedure using exo complex 11b ( $0.53 \mathrm{~g}, 1.2 \mathrm{mmol}$ ). The alcohol 12b ( $293 \mathrm{mg}, 66 \%$ over three steps) was obtained as a colourless oil; $[a]_{\mathrm{D}}^{31}+7.7$ ( $c 1.00$ in $\mathrm{CHCl}_{3}$ ) (Found: C, 63.3; H, 12.0. $\mathrm{C}_{22} \mathrm{H}_{50} \mathrm{O}_{3} \mathrm{Si}_{2}$ requires C, 63.1; $\mathrm{H}, 12.0 \%) ; v_{\max }(\mathrm{film}) / \mathrm{cm}^{-1} 3344(\mathrm{OH}), 2954,2930,2857,1472$, $1462,1375,1361,1255,1130,1049,1005,939,835,773 ; \delta_{\mathrm{H}}(600$ MHz) $3.82-3.73(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}$ or $9-\mathrm{H}), 3.68-3.62(3 \mathrm{H}, \mathrm{m}[\mathrm{incl}$. $3.64,2 \mathrm{H}, \mathrm{t}, J 6.5,2 \times 1-\mathrm{H}] 2 \times 1-\mathrm{H}, 9-\mathrm{H}$ or $5-\mathrm{H}), 1.68-1.20$ $(13 \mathrm{H}, \mathrm{m}, 2 \times 2-\mathrm{H}, 2 \times 3-\mathrm{H}, 2 \times 4-\mathrm{H}, 2 \times 6-\mathrm{H}, 2 \times 7-\mathrm{H}, 2 \times 8-\mathrm{H}$, $\mathrm{OH}), 1.11(3 \mathrm{H}, \mathrm{d}, J 6.1,3 \times 10-\mathrm{H}), 0.99\left(9 \mathrm{H}, \mathrm{s}, \mathrm{SiC}\left(\mathrm{CH}_{3}\right)_{3}\right), 0.88$ $\left(9 \mathrm{H}, \mathrm{s}, \mathrm{SiC}\left(\mathrm{CH}_{3}\right)_{3}\right), 0.07-0.03\left(12 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{Si}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta_{\mathrm{C}}(150$ MHz) $72.2(\mathrm{CH}), 68.6(\mathrm{CH}), 63.0(1-\mathrm{C}), 40.1\left(\mathrm{CH}_{2}\right), 37.2\left(\mathrm{CH}_{2}\right)$, $36.7\left(\mathrm{CH}_{2}\right), 33.0\left(\mathrm{CH}_{2}\right), 25.9\left(2 \times \mathrm{SiC}\left(\mathrm{CH}_{3}\right)_{3}\right), 23.8(10-\mathrm{C})$, $21.6\left(\mathrm{CH}_{2}\right), 21.3\left(\mathrm{CH}_{2}\right), 18.11\left(\mathrm{SiC}\left(\mathrm{CH}_{3}\right)_{3}\right), 18.10\left(\mathrm{Si} C\left(\mathrm{CH}_{3}\right)_{3}\right)$, $\left.-4.4\left(2 \times \mathrm{SiCH}_{3}\right),-4.7(\mathrm{SiCH})_{3}\right) ; m / z(\mathrm{FIB}) 419\left(\mathrm{MH}^{+}, 8 \%\right)$, 345 (12), 287 (20), 285 (20), 229 (31), 185 (40), 159 (46), 155 (100), 137 (71), 133 (68), 115 (65) [Found $\left(\mathrm{MH}^{+}\right) 419.3363$. $\mathrm{C}_{22} \mathrm{H}_{51} \mathrm{O}_{3} \mathrm{Si}_{2}$ requires $\mathrm{MH}, 419.3377$ ].

## General procedure for the preparation of aldehydes 13a and 13b

For a 0.70 mmol scale reaction: a solution of DMSO (2.6 equiv.) in DCM ( 1.5 ml ) was added dropwise to a solution of $(\mathrm{COCl})_{2}(1.3$ equiv. $)$ in $\mathrm{DCM}(2.0 \mathrm{ml})$ at $-78^{\circ} \mathrm{C}$ and the reaction mixture was stirred for 30 min . A solution of the alcohol ( 1.0 equiv.) in DCM ( 1.5 ml ) was added dropwise and after $30 \mathrm{~min} \mathrm{Et}_{3} \mathrm{~N}$ ( 3.5 equiv.) was added dropwise at $-78^{\circ} \mathrm{C}$. Stirring was continued at $-78^{\circ} \mathrm{C}$ for 1 h before the mixture was allowed to warm to rt . The reaction mixture was poured into water $(10 \mathrm{ml})$ and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 10 \mathrm{ml})$. The combined organic extracts were washed with brine ( 5 ml ), dried over $\mathrm{MgSO}_{4}$ and concentrated under reduced pressure. Purification of the residue by flash column chromatography (silica gel, $10 \%$ $\mathrm{Et}_{2} \mathrm{O}$ in PE ) afforded the aldehyde.
(5S,9S)-5,9-Bis(tert-butyldimethylsilyloxy)decanal (13a). Prepared according to the general procedure using alcohol 12a ( $253 \mathrm{mg}, 0.61 \mathrm{mmol}$ ). The aldehyde 13a ( $218 \mathrm{mg}, 86 \%$ ) was obtained as a colourless liquid; $[a]_{\mathrm{D}}^{31}+10.6\left(c 1.00\right.$ in $\left.\mathrm{CHCl}_{3}\right)$
(Found: $\mathrm{C}, 63.7 ; \mathrm{H}, 11.5 . \mathrm{C}_{22} \mathrm{H}_{48} \mathrm{O}_{3} \mathrm{Si}_{2}$ requires $\mathrm{C}, 63.4 ; \mathrm{H}$, $11.6 \%)$; $v_{\text {max }}($ film $) / \mathrm{cm}^{-1} 2953,2929,2894,2857,2710(\mathrm{O}=\mathrm{C}-\mathrm{H})$, $1729(\mathrm{C}=\mathrm{O}), 1472,1462,1375,1360,1255,1129,1073,1048$, $1005,835,774 ; \delta_{\mathrm{H}}(200 \mathrm{MHz}) 9.74(1 \mathrm{H}, \mathrm{t}, J 1.8,1-\mathrm{H}), 3.85-3.55$ $(2 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}, 9-\mathrm{H}), 2.40(2 \mathrm{H}, \mathrm{dt}, J 7.1,1.8,2 \times 2-\mathrm{H}), 1.85-1.15$ $(10 \mathrm{H}, \mathrm{m}, 2 \times 3-\mathrm{H}, 2 \times 4-\mathrm{H}, 2 \times 6-\mathrm{H}, 2 \times 7-\mathrm{H}, 2 \times 8-\mathrm{H}), 1.09$ $(3 \mathrm{H}, \mathrm{d}, J 6.1,3 \times 10-\mathrm{H}), 0.87\left(18 \mathrm{H}, \mathrm{s}, 2 \times \mathrm{SiC}\left(\mathrm{CH}_{3}\right)_{3}\right), 0.02$ $\left(12 \mathrm{H}, \mathrm{s}, 2 \times \mathrm{Si}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta_{\mathrm{C}}(50 \mathrm{MHz}) 202.5(1-\mathrm{C}), 71.7(5-\mathrm{C}$ or 9-C), $68.5(9-\mathrm{C}$ or $5-\mathrm{C}), 43.9\left(\mathrm{CH}_{2}\right), 39.9\left(\mathrm{CH}_{2}\right), 37.0\left(\mathrm{CH}_{2}\right)$, $36.2\left(\mathrm{CH}_{2}\right), 25.8\left(2 \times \mathrm{SiC}\left(\mathrm{CH}_{3}\right)_{3}\right), 23.7(10-\mathrm{C}), 21.3\left(\mathrm{CH}_{2}\right), 18.0$ $\left(2 \times \mathrm{SiC}\left(\mathrm{CH}_{3}\right)_{3}\right), 17.8\left(\mathrm{CH}_{2}\right),-4.5(2 \times \mathrm{SiCH} 3),-4.6\left(\mathrm{SiCH}_{3}\right)$, $-4.8\left(\mathrm{SiCH}_{3}\right) ; m / z(\mathrm{FIB}) 415\left(\mathrm{M}^{+}-\mathrm{H}, 5 \%\right), 401$ (13), 399 $\left(\mathrm{M}^{+}-\mathrm{OH}, 2\right), 359\left(\mathrm{M}^{+}-{ }^{t} \mathrm{Bu}, 68\right), 285$ (12), 227 (71), 185 (100) [Found $\left(\mathrm{M}^{+}-\mathrm{H}\right)$ 415.3083. $\mathrm{C}_{22} \mathrm{H}_{47} \mathrm{O}_{3} \mathrm{Si}_{2}$ requires $\mathrm{M}-\mathrm{H}$, 415.3064].
(5R,9S)-5,9-Bis(tert-butyldimethylsilyloxy)decanal (13b). Prepared according to the general procedure using alcohol 12b $(340 \mathrm{mg}, 0.81 \mathrm{mmol})$. The aldehyde 13b ( $269 \mathrm{mg}, 80 \%$ ) was obtained as a colourless liquid; $[a]_{\mathrm{D}}^{31}+5.8\left(c 1.00\right.$ in $\left.\mathrm{CHCl}_{3}\right)$ (Found: C , 63.5; $\mathrm{H}, 11.6 . \mathrm{C}_{22} \mathrm{H}_{48} \mathrm{O}_{3} \mathrm{Si}_{2}$ requires $\mathrm{C}, 63.4 ; \mathrm{H}$, $11.6 \%) ; v_{\max }(\mathrm{film}) / \mathrm{cm}^{-1} 2954,2930,2900,2857,2710(\mathrm{O}=\mathrm{C}-\mathrm{H})$, $1730(\mathrm{C}=\mathrm{O}), 1472,1462,1375,1360,1255,1129,1047,1005$, 835,$773 ; \delta_{\mathrm{H}}(200 \mathrm{MHz}) 9.75(1 \mathrm{H}, \mathrm{t}, J 1.8,1-\mathrm{H}), 3.87-3.55(2 \mathrm{H}$, $\mathrm{m}, 5-\mathrm{H}, 9-\mathrm{H}), 2.41(2 \mathrm{H}, \mathrm{dt}, J 7.1,1.8,2 \times 2-\mathrm{H}), 1.85-1.20(10 \mathrm{H}$, $\mathrm{m}, 2 \times 3-\mathrm{H}, 2 \times 4-\mathrm{H}, 2 \times 6-\mathrm{H}, 2 \times 7-\mathrm{H}, 2 \times 8-\mathrm{H}), 1.10(3 \mathrm{H}, \mathrm{d}$, $J 6.1,3 \times 10-\mathrm{H}), 0.88\left(18 \mathrm{H}, \mathrm{s}, 2 \times \mathrm{SiC}\left(\mathrm{CH}_{3}\right)_{3}\right), 0.03(12 \mathrm{H}, \mathrm{s}$, $\left.2 \times \mathrm{Si}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta_{\mathrm{C}}(150 \mathrm{MHz}) 202.4(1-\mathrm{C}), 71.8(5-\mathrm{C}$ or $9-\mathrm{C}), 68.5$ (9-C or 5-C), $44.0\left(\mathrm{CH}_{2}\right), 40.0\left(\mathrm{CH}_{2}\right), 37.1\left(\mathrm{CH}_{2}\right), 36.2\left(\mathrm{CH}_{2}\right)$, $25.8\left(2 \times \mathrm{SiC}\left(\mathrm{CH}_{3}\right)_{3}\right), 23.7(10-\mathrm{C}), 21.5\left(\mathrm{CH}_{2}\right), 18.1(\mathrm{SiC}-$ $\left.\left(\mathrm{CH}_{3}\right)_{3}\right), 18.0\left(\mathrm{SiC}\left(\mathrm{CH}_{3}\right)_{3}\right), 17.8\left(\mathrm{CH}_{2}\right),-4.5\left(2 \times \mathrm{SiCH}_{3}\right),-4.6$ $\left(\mathrm{SiCH}_{3}\right),-4.8\left(\mathrm{SiCH}_{3}\right) ; m / z(\mathrm{FIB}) 415\left(\mathrm{M}^{+}-\mathrm{H}, 4 \%\right), 359$ $\left(\mathrm{M}^{+}-{ }^{t} \mathrm{Bu}, 62\right), 227$ (68), 185 (100), 149 (75) [Found ( $\left.\mathrm{M}^{+}-\mathrm{H}\right)$ 415.3075. $\mathrm{C}_{22} \mathrm{H}_{47} \mathrm{O}_{3} \mathrm{Si}_{2}$ requires $\mathrm{M}-\mathrm{H}, 415.3064$ ].

## General procedure for the preparation of stannanes 14a and 14b

For a 0.50 mmol scale reaction: a solution of tributyldiiodomethylstannane ${ }^{14}$ ( 2.0 equiv.) and the aldehyde ( 1.0 equiv.) in DMF $(10 \mathrm{ml})$ was added to a solution of $\mathrm{CrCl}_{2}(10.0$ equiv.) in DMF ( 30 ml ) and the reaction mixture was stirred for 5 h . The reaction mixture was carefully poured into ice-water $(100 \mathrm{ml})$ and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 50 \mathrm{ml})$. The combined organic extracts were washed with water $(30 \mathrm{ml})$ and brine $(30 \mathrm{ml})$, dried over $\mathrm{MgSO}_{4}$ and concentrated under reduced pressure. Purification of the residue by flash column chromatography (silica gel $100 \mathrm{C}_{18}$-reversed phase, $30 \% \mathrm{DCM}$ in $\mathrm{MeOH})$ afforded the stannane ( $c a .19: 1$ mixture of $E-Z$ isomers).
( $1 E, 6 R, 10 S$ )-6,10-Bis(tert-butyldimethylsilyloxy)-1-tributyl-stannylundec-1-ene (14a). Prepared according to the general procedure using aldehyde 13a ( $216 \mathrm{mg}, 0.52 \mathrm{mmol}$ ). The stannane 14a ( $248 \mathrm{mg}, 67 \%$ ) was obtained as a colourless liquid; $v_{\text {max }}($ film $) / \mathrm{cm}^{-1} 2956,2928,2956,1598(\mathrm{C}=\mathrm{C}), 1462,1376,1254$, $1048,835,773$; NMR data quoted only for $E$-isomer: $\delta_{\mathrm{H}}(400$ $\mathrm{MHz}) 6.15-5.55(2 \mathrm{H}, \mathrm{m}, 1-\mathrm{H}, 2-\mathrm{H}), 3.84-3.73(1 \mathrm{H}, \mathrm{m}, 10-\mathrm{H})$, $3.70-3.58(1 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}), 2.18-2.06(2 \mathrm{H}, \mathrm{m}, 2 \times 3-\mathrm{H}), 1.65-1.20$ $(22 \mathrm{H}, \mathrm{m}, 2 \times 4-\mathrm{H}, 2 \times 5-\mathrm{H}, 2 \times 7-\mathrm{H}, 2 \times 8-\mathrm{H}, 2 \times 9-\mathrm{H}, 3 \times$ $\left.\mathrm{Sn}\left(\mathrm{CH}_{2}\right)_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.12(3 \mathrm{H}, \mathrm{d}, J 6.1,3 \times 11-\mathrm{H}), 0.96-0.83$ $\left(33 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{SiC}\left(\mathrm{CH}_{3}\right)_{3}, 3 \times \mathrm{Sn}\left(\mathrm{CH}_{2}\right)_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 0.5(6 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{Si}\left(\mathrm{CH}_{3}\right)_{2}\right), 0.04\left(6 \mathrm{H}, \mathrm{s}, \mathrm{Si}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta_{\mathrm{C}}(100 \mathrm{MHz}) 149.5(1-\mathrm{C}$ or $2-\mathrm{C}), 127.2$ ( $2-\mathrm{C}$ or $1-\mathrm{C}$ ), 72.2 ( $6-\mathrm{C}$ or $10-\mathrm{C}$ ), 68.6 ( $10-\mathrm{C}$ or $6-\mathrm{C}$ ), $40.0\left(\mathrm{CH}_{2}\right), 37.9\left(\mathrm{CH}_{2}\right), 37.2\left(\mathrm{CH}_{2}\right), 36.6\left(\mathrm{CH}_{2}\right), 29.1\left(3 \times \mathrm{CH}_{2}\right)$, $27.2\left(3 \times \mathrm{CH}_{2}\right), 25.9\left(2 \times \mathrm{SiC}\left(\mathrm{CH}_{3}\right)_{3}\right), 24.6\left(\mathrm{CH}_{2}\right), 23.7(11-\mathrm{C})$, $21.5\left(\mathrm{CH}_{2}\right), 18.1\left(2 \times \mathrm{SiC}\left(\mathrm{CH}_{3}\right)_{3}\right), 13.7\left(\mathrm{Sn}\left[\left(\mathrm{CH}_{2}\right)_{3} \mathrm{CH}_{3}\right]_{3}\right), 9.4$ $\left(3 \times \mathrm{CH}_{2}\right),-4.4\left(2 \times \mathrm{SiCH} \mathrm{H}_{3}\right),-4.7(2 \times \mathrm{SiCH})^{2}$.
(1E,6S,10S)-6,10-Bis(tert-butyldimethylsilyloxy)-1-tributyl-stannylundec-1-ene (14b). Prepared according to the general procedure using aldehyde $\mathbf{1 3 b}(400 \mathrm{mg}, 0.96 \mathrm{mmol})$. The
stannane 14b ( $467 \mathrm{mg}, 69 \%$ ) was obtained as a colourless liquid; $v_{\text {max }}($ film $) / \mathrm{cm}^{-1} 2956,2928,2866,1599(\mathrm{C}=\mathrm{C}), 1463,1376,1360$, $1264,1048,836,773$; NMR data quoted only for $E$-isomer: $\delta_{\mathrm{H}}(400 \mathrm{MHz}) 6.10-5.35(2 \mathrm{H}, \mathrm{m}, 1-\mathrm{H}, 2-\mathrm{H}), 3.85-3.72(1 \mathrm{H}, \mathrm{m}$, $10-\mathrm{H}), 3.70-3.57(1 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}), 2.19-2.06(2 \mathrm{H}, \mathrm{m}, 2 \times 3-\mathrm{H})$, $1.60-1.24(22 \mathrm{H}, \mathrm{m}, 2 \times 4-\mathrm{H}, 2 \times 5-\mathrm{H}, 2 \times 7-\mathrm{H}, 2 \times 8-\mathrm{H}, 2 \times 9-\mathrm{H}$, $\left.3 \times \mathrm{Sn}\left(\mathrm{CH}_{2}\right)_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.11(3 \mathrm{H}, \mathrm{d}, \mathrm{J} 6.0,3 \times 11-\mathrm{H}), 0.94-0.84$ $\left(33 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{SiC}\left(\mathrm{CH}_{3}\right)_{3}, 3 \times \mathrm{Sn}\left(\mathrm{CH}_{2}\right)_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 0.05(6 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{Si}\left(\mathrm{CH}_{3}\right)_{2}\right), 0.04\left(6 \mathrm{H}, \mathrm{s}, \mathrm{Si}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta_{\mathrm{C}}(100 \mathrm{MHz}) 149.5(1-\mathrm{C}$ or $2-\mathrm{C}), 127.2$ ( $2-\mathrm{C}$ or $1-\mathrm{C}$ ), 72.2 ( $6-\mathrm{C}$ or $10-\mathrm{C}$ ), 68.6 ( $10-\mathrm{C}$ or $6-\mathrm{C}$ ), $40.1\left(\mathrm{CH}_{2}\right), 37.9\left(\mathrm{CH}_{2}\right), 37.3\left(\mathrm{CH}_{2}\right), 36.5\left(\mathrm{CH}_{2}\right), 29.1\left(3 \times \mathrm{CH}_{2}\right)$, $27.2\left(3 \times \mathrm{CH}_{2}\right), 25.9\left(2 \times \mathrm{SiC}\left(\mathrm{CH}_{3}\right)_{3}\right), 24.5\left(\mathrm{CH}_{2}\right), 23.8(11-\mathrm{C})$, $21.6\left(\mathrm{CH}_{2}\right), 18.1\left(2 \times \mathrm{SiC}\left(\mathrm{CH}_{3}\right)_{3}\right), 13.7\left(\mathrm{Sn}\left[\left(\mathrm{CH}_{2}\right)_{3} \mathrm{CH}_{3}\right]_{3}\right), 9.4$ $\left(3 \times \mathrm{CH}_{2}\right),-4.4\left(2 \times \mathrm{SiCH}_{3}\right),-4.7\left(2 \times \mathrm{SiCH}_{3}\right)$.

## General procedure for the Stille coupling of the stannanes 14 a and 14b

For a 0.50 mmol scale reaction: $\mathrm{Pd}_{2}(\mathrm{dba})_{3}(0.05$ equiv.) and $\mathrm{P}(2 \text {-furyl })_{3}$ ( 0.4 equiv.) were dissolved in $\mathrm{PhMe}(1 \mathrm{ml})$ and stirred at rt until the solution had turned from purple to yellow. This solution was added to a mixture of the stannane (1.0 equiv.) and the aryl iodide $15^{6 f}$ ( 1.0 equiv.) and the reaction was heated to $80^{\circ} \mathrm{C}$ for 4 h . The reaction mixture was poured into aqueous $\mathrm{NH}_{4} \mathrm{Cl}$ solution $(10 \mathrm{ml})$ and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times$ $10 \mathrm{ml})$. The combined organic extracts were washed with brine $(5 \mathrm{ml})$, dried over $\mathrm{MgSO}_{4}$ and concentrated under reduced pressure. Purification of the residue by flash column chromatography (silica gel, $70 \% \mathrm{Et}_{2} \mathrm{O}$ in PE ) afforded the coupled product.

Methyl 4,6-bis[(2-methoxyethoxy)methyloxy]-2-[(1'E,6'R, $\left.10^{\prime} S\right)-\mathbf{6}^{\prime}, 10^{\prime}$-bis(tert-butyldimethylsilyloxy)undec-1' -en-1'-yl]-
benzoate (16a). Prepared according to the general procedure using stannane $\mathbf{1 4 a}$ ( $352 \mathrm{mg}, 0.50 \mathrm{mmol}$ ). The coupled product 16a ( $310 \mathrm{mg}, 82 \%$ ) was obtained as a colourless liquid; $[\alpha]_{\mathrm{D}}^{30}$ $+5.1\left(c 1.00\right.$ in $\left.\mathrm{CHCl}_{3}\right) ; v_{\max }($ film $) / \mathrm{cm}^{-1} 2956,2925,2855,1733$ $(\mathrm{C}=\mathrm{O}), 1602(\mathrm{C}=\mathrm{C}), 1456,1254,1107,1024,835,772 ; \delta_{\mathrm{H}}(400$ $\mathrm{MHz}) 6.84(1 \mathrm{H}, \mathrm{d}, J 2.0, \mathrm{Ar}-H), 6.73(1 \mathrm{H}, \mathrm{d}, J 2.0, \mathrm{Ar}-H), 6.30$ $\left(1 \mathrm{H}, \mathrm{d}, J 15.7,1^{\prime}-\mathrm{H}\right), 6.16\left(1 \mathrm{H}, \mathrm{dt}, J 15.7,6.7,2^{\prime}-\mathrm{H}\right), 5.25(2 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{OCH}_{2} \mathrm{O}\right), 5.23\left(2 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{2} \mathrm{O}\right), 3.87\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 3.84-$ $3.74\left(5 \mathrm{H}, \mathrm{m}, 10^{\prime}-\mathrm{H}, 2 \times \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 3.68-3.60\left(1 \mathrm{H}, \mathrm{m}, 6^{\prime}-\mathrm{H}\right)$, 3.58-3.51 (4H, m, $\left.2 \times \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 3.39\left(3 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{3}\right), 3.36$ $\left(3 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{3}\right), 2.20-2.10\left(2 \mathrm{H}, \mathrm{m}, 2 \times 3^{\prime}-\mathrm{H}\right), 1.55-1.20(10 \mathrm{H}, \mathrm{m}$, $\left.2 \times 4^{\prime}-\mathrm{H}, 2 \times 5^{\prime}-\mathrm{H}, 2 \times 7^{\prime}-\mathrm{H}, 2 \times 8^{\prime}-\mathrm{H}, 2 \times 9^{\prime}-\mathrm{H}\right), 1.07(3 \mathrm{H}, \mathrm{d}$, $\left.J 6.1,3 \times 11^{\prime}-\mathrm{H}\right), 0.88\left(18 \mathrm{H}, \mathrm{s}, 2 \times \mathrm{SiC}\left(\mathrm{CH}_{3}\right)_{3}\right), 0.04(12 \mathrm{H}, \mathrm{s}$, $\left.2 \times \mathrm{Si}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta_{\mathrm{C}}(100 \mathrm{MHz}) 168.3\left(\mathrm{CO}_{2} \mathrm{CH}_{3}\right), 158.8$, (quat. Ar-C), 155.3 (quat. Ar-C), 137.9 (quat. Ar-C), 134.7 (2'-C), 126.2 ( $1^{\prime}-\mathrm{C}$ ), 117.4 (quat. Ar-C), 106.1 (Ar-C), 102.5 (Ar-C), $93.8\left(\mathrm{OCH}_{2} \mathrm{O}\right), 93.3\left(\mathrm{OCH}_{2} \mathrm{O}\right), 72.1\left(6^{\prime}-\mathrm{C}\right), 71.5\left(2 \times \mathrm{OCH}_{2}-\right.$ $\left.\mathrm{CH}_{2} \mathrm{O}\right), 68.6\left(10^{\prime}-\mathrm{C}\right), 67.8\left(\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 67.7\left(\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right)$, $59.0\left(2 \times \mathrm{OCH}_{3}\right)$, $52.1\left(\mathrm{CO}_{2} \mathrm{CH}_{3}\right), 40.0\left(\mathrm{CH}_{2}\right), 37.3\left(\mathrm{CH}_{2}\right)$, $36.7\left(\mathrm{CH}_{2}\right), 33.4\left(3^{\prime}-\mathrm{C}\right), 25.9\left(2 \times \mathrm{SiC}\left(\mathrm{CH}_{3}\right)_{3}\right), 24.9\left(\mathrm{CH}_{2}\right), 23.7$ $\left(11^{\prime}-\mathrm{C}\right), 21.5\left(\mathrm{CH}_{2}\right), 18.1\left(2 \times \mathrm{SiC}\left(\mathrm{CH}_{3}\right)_{3}\right),-4.4\left(2 \times \mathrm{SiCH}_{3}\right)$, $-4.7\left(2 \times \mathrm{SiCH}_{3}\right) ; m / z(\mathrm{ES}) 779\left(\mathrm{MNa}^{+}, 6 \%\right)$, 734 (4), 597 (100), 539 (16) [Found $\left(\mathrm{MNa}^{+}\right)$779.4534. $\mathrm{C}_{39} \mathrm{H}_{72} \mathrm{NaO}_{10} \mathrm{Si}_{2}$ requires $\mathrm{MNa}, 779.4562$ ].

Methyl 4,6-bis[(2-methoxyethoxy)methyloxy]-2-[(1'E,6'S, $10^{\prime} S$ )-6', $\mathbf{1 0}^{\prime}$-bis(tert-butyldimethylsilyloxy)undec-1'-en-1'-yl]benzoate (16b). Prepared according to the general procedure using stannane $\mathbf{1 4 b}$ ( $317 \mathrm{mg}, 0.45 \mathrm{mmol}$ ). The coupled product 16b ( $290 \mathrm{mg}, 85 \%$ ) was obtained as a colourless liquid; $[\alpha]_{\mathrm{D}}^{30}$ $+4.4\left(c 1.00\right.$ in $\left.\mathrm{CHCl}_{3}\right) ; v_{\max }($ (film $) / \mathrm{cm}^{-1} 2951,2928,2856,1732$ $(\mathrm{C}=\mathrm{O}), 1601(\mathrm{C}=\mathrm{C}), 1578(\mathrm{C}=\mathrm{C}), 1472,1462,1431,1256,1108$, $1022,835,773 ; \delta_{\mathrm{H}}(600 \mathrm{MHz}) 6.84(1 \mathrm{H}, \mathrm{d}, J 1.8, \mathrm{Ar}-H), 6.73$ $(1 \mathrm{H}, \mathrm{d}, J 1.8, \operatorname{Ar}-H), 6.31\left(1 \mathrm{H}, \mathrm{d}, J 15.7,1^{\prime}-\mathrm{H}\right), 6.17(1 \mathrm{H}, \mathrm{dt}$, $\left.J 15.7,6.8,2^{\prime}-\mathrm{H}\right), 5.26\left(2 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{2} \mathrm{O}\right), 5.24\left(2 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{2} \mathrm{O}\right)$, $3.88\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 3.85-3.72\left(5 \mathrm{H}, \mathrm{m}, 10^{\prime}-\mathrm{H}, 2 \times \mathrm{OCH}_{2}-\right.$ $\left.\mathrm{CH}_{2} \mathrm{O}\right), 3.70-3.62\left(1 \mathrm{H}, \mathrm{m}, 6^{\prime}-\mathrm{H}\right), 3.58-3.51\left(4 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{OCH}_{2}-\right.$
$\left.\mathrm{CH}_{2} \mathrm{O}\right), 3.38\left(3 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{3}\right), 3.37\left(3 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{3}\right), 2.22-2.12(2 \mathrm{H}$, $\left.\mathrm{m}, 2 \times 3^{\prime}-\mathrm{H}\right), 1.55-1.23\left(10 \mathrm{H}, \mathrm{m}, 2 \times 4^{\prime}-\mathrm{H}, 2 \times 5^{\prime}-\mathrm{H}, 2 \times 7^{\prime}-\mathrm{H}\right.$, $\left.2 \times 8^{\prime}-\mathrm{H}, 2 \times 9^{\prime}-\mathrm{H}\right), 1.11\left(3 \mathrm{H}, \mathrm{d}, J 6.1,3 \times 11^{\prime}-\mathrm{H}\right), 0.88(18 \mathrm{H}, \mathrm{s}$, $\left.2 \times \mathrm{SiC}\left(\mathrm{CH}_{3}\right)_{3}\right), 0.04\left(12 \mathrm{H}, \mathrm{s}, 2 \times \mathrm{Si}\left(\mathrm{CH}_{3}\right)_{2}\right) ; \delta_{\mathrm{C}}(150 \mathrm{MHz}) 168.3$ $\left(\mathrm{CO}_{2} \mathrm{CH}_{3}\right), 158.8$ (quat. Ar-C), 155.2 (quat. Ar-C), 137.9 (quat Ar-C), 134.7 ( $2^{\prime}-\mathrm{C}$ ), 126.2 ( $\left.1^{\prime}-\mathrm{C}\right), 117.4$ (quat. Ar-C), 106.1 (Ar-C), $102.5(\mathrm{Ar}-\mathrm{C}), 93.8\left(\mathrm{OCH}_{2} \mathrm{O}\right), 93.3\left(\mathrm{OCH}_{2} \mathrm{O}\right), 72.1$ ( $\left.6^{\prime}-\mathrm{C}\right), 71.5\left(2 \times \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 68.6\left(10^{\prime}-\mathrm{C}\right), 67.8\left(\mathrm{OCH}_{2}-\right.$ $\left.\mathrm{CH}_{2} \mathrm{O}\right), 67.7\left(\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 59.0\left(\mathrm{OCH}_{3}\right), 58.9\left(\mathrm{OCH}_{3}\right), 52.1$ $\left(\mathrm{CO}_{2} \mathrm{CH}_{3}\right), 40.1\left(\mathrm{CH}_{2}\right), 37.3\left(\mathrm{CH}_{2}\right), 36.6\left(\mathrm{CH}_{2}\right), 33.4\left(3^{\prime}-\mathrm{C}\right)$, $25.9\left(2 \times \mathrm{SiC}\left(\mathrm{CH}_{3}\right)_{3}\right), 24.8\left(\mathrm{CH}_{2}\right), 23.7\left(11^{\prime}-\mathrm{C}\right), 21.6\left(\mathrm{CH}_{2}\right), 18.1$ $\left(2 \times \mathrm{SiC}\left(\mathrm{CH}_{3}\right)_{3}\right),-4.4\left(2 \times \mathrm{SiCH} \mathrm{H}_{3}\right),-4.7\left(2 \times \mathrm{SiCH}_{3}\right) ; m / z$ (FAB) $779\left(\mathrm{MNa}^{+}, 100 \%\right), 756\left(\mathrm{M}^{+}, 17\right), 725\left(\mathrm{M}^{+}-\mathrm{CH}_{3} \mathrm{O}\right)$, 12), $699\left(\mathrm{M}^{+}-{ }^{'} \mathrm{Bu}, 59\right), 681$ (93) [Found $\left(\mathrm{MNa}^{+}\right) 779.4616$. $\mathrm{C}_{39} \mathrm{H}_{72} \mathrm{NaO}_{10} \mathrm{Si}_{2}$ requires $\left.\mathrm{MNa}, 779.4562\right]$.

## Preparation of stock solution of HF-py in pyridine and THF

HF•py ( 3.8 ml ) was added carefully to a stirred solution of pyridine ( 14 ml ) in THF ( 40 ml ) in a polyvinylchloride bottle under argon. The resulting colourless solution was stored under argon at $-20^{\circ} \mathrm{C}$ and used as the stock solution in the following deprotections.

## General procedure for the TBDMS-deprotection of bis-silyl ethers 16a and 16b

For a 0.10 mmol scale reaction: the bis-silyl ether ( 1.0 equiv.) was treated with HF-py stock solution ( 3 ml ) at rt for 12 h . The reaction mixture was poured into aqueous $\mathrm{CuSO}_{4}$ solution $(10 \mathrm{ml})$ and extracted with $\mathrm{EtOAc}(3 \times 10 \mathrm{ml})$. The combined organic extracts were dried over $\mathrm{MgSO}_{4}$ and concentrated under reduced pressure. Purification of the residue by flash column chromatography (silica gel, $5 \% \mathrm{MeOH}$ in $\mathrm{Et}_{2} \mathrm{O}$ ) afforded the diol.

Methyl 4,6-bis[(2-methoxyethoxy)methyloxy]-2-[(1' $\boldsymbol{1}^{\prime} \mathbf{6}^{\prime} R$, $\mathbf{1 0}^{\prime} \boldsymbol{S}$ )- $\mathbf{6}^{\prime}, \mathbf{1 0}^{\prime}$-dihydroxyundec-1'-en-1'-yl]benzoate (17a). Prepared according to the general procedure using bis-silyl ether 16a ( $83 \mathrm{mg}, 0.11 \mathrm{mmol}$ ). The diol 17a ( $55 \mathrm{mg}, 95 \%$ ) was obtained as a colourless liquid; $[a]_{\mathrm{D}}^{29}+3.5$ ( c 1.30 in $\mathrm{CHCl}_{3}$ ); $v_{\max }(\mathrm{film}) / \mathrm{cm}^{-1} 3416(\mathrm{OH}), 2928,1729$ (C=O), 1601 (C=C), 1577 (C=C), 1433, 1267, 1157, 1108, 1020, 849; $\delta_{\mathrm{H}}(400 \mathrm{MHz}) 6.84$ $(1 \mathrm{H}, \mathrm{d}, J 2.0, \operatorname{Ar}-H), 6.73(1 \mathrm{H}, \mathrm{d}, J 2.0, \operatorname{Ar}-H), 6.32(1 \mathrm{H}, \mathrm{d}$, $J$ 15.7, $\left.1^{\prime}-\mathrm{H}\right), 6.21-6.10\left(1 \mathrm{H}, \mathrm{m}, 2^{\prime}-\mathrm{H}\right), 5.26\left(2 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{2} \mathrm{O}\right)$, $5.23\left(2 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{2} \mathrm{O}\right), 3.88\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 3.85-3.76(5 \mathrm{H}, \mathrm{m}$, $10^{\prime}-\mathrm{H}, 2 \times \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}$ ), $3.67-3.58\left(1 \mathrm{H}, \mathrm{m}, 6^{\prime}-\mathrm{H}\right), 3.58-3.51$ $\left(4 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 3.37\left(3 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{3}\right), 3.36(3 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{OCH}_{3}\right), 2.24-2.14\left(2 \mathrm{H}, \mathrm{m}, 2 \times 3^{\prime}-\mathrm{H}\right), 1.70-1.30(10 \mathrm{H}, \mathrm{m}, 2 \times$ $\left.4^{\prime}-\mathrm{H}, 2 \times 5^{\prime}-\mathrm{H}, 2 \times 7^{\prime}-\mathrm{H}, 2 \times 8^{\prime}-\mathrm{H}, 2 \times 9^{\prime}-\mathrm{H}\right), 1.18(3 \mathrm{H}, \mathrm{d}, J 6.2$, $\left.3 \times 11^{\prime}-\mathrm{H}\right) ; \delta_{\mathrm{C}}(100 \mathrm{MHz}) 168.4\left(\mathrm{CO}_{2} \mathrm{CH}_{3}\right), 158.8$, (quat. Ar-C), 155.3 (quat. Ar-C), 137.8 (quat. Ar-C), 134.4 (2'-C), 126.6 ( $1^{\prime}-\mathrm{C}$ ), 117.3 (quat. $\mathrm{Ar}-\mathrm{C}$ ), 106.0 ( $\mathrm{Ar}-\mathrm{C}$ ), 102.5 ( $\mathrm{Ar}-\mathrm{C}$ ), 93.8 $\left(\mathrm{OCH}_{2} \mathrm{O}\right), 93.3\left(\mathrm{OCH}_{2} \mathrm{O}\right), 71.6\left(6^{\prime}-\mathrm{C}\right), 71.5\left(2 \times \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right)$, $68.0\left(10^{\prime}-\mathrm{C}\right), 67.8\left(\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 67.7\left(\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 59.0(2 \times$ $\left.\mathrm{OCH}_{3}\right), 52.2\left(\mathrm{CO}_{2} \mathrm{CH}_{3}\right), 39.2\left(\mathrm{CH}_{2}\right), 37.3\left(\mathrm{CH}_{2}\right), 36.8\left(\mathrm{CH}_{2}\right)$, $32.9\left(3^{\prime}-\mathrm{C}\right), 25.0\left(\mathrm{CH}_{2}\right), 23.5\left(11^{\prime}-\mathrm{C}\right), 21.7\left(\mathrm{CH}_{2}\right) ; m / z(\mathrm{ES}) 551$ $\left(\mathrm{MNa}^{+}, 100 \%\right)$ [Found ( $\mathrm{MNa}^{+}$) 551.2793. $\mathrm{C}_{27} \mathrm{H}_{44} \mathrm{NaO}_{10}$ requires MNa , 551.2832].

## Methyl 4,6-bis[(2-methoxyethoxy)methyloxy]-2-[(1' $E, 6^{\prime} S$,

 $10^{\prime} S$ )- $\mathbf{6}^{\prime}, 10^{\prime}$-dihydroxyundec-1'-en- $\mathbf{1}^{\prime}$-yl]benzoate ( $\mathbf{1 7 b}$ ). Prepared according to the general procedure using bis-silyl ether 16b ( $68 \mathrm{mg}, 0.09 \mathrm{mmol}$ ). The diol 17b ( $44 \mathrm{mg}, 93 \%$ ) was obtained as a colourless liquid; $[\alpha]_{D}^{33}+5.4$ (c 1.00 in $\mathrm{CHCl}_{3}$ ); $v_{\max }(\mathrm{film}) / \mathrm{cm}^{-1} 3395(\mathrm{OH}), 2930,1728(\mathrm{C}=\mathrm{O}), 1600(\mathrm{C}=\mathrm{C}), 1578$ (C=C), 1433, 1268, 1157, 1109, 1020, 849; $\delta_{\mathrm{H}}(400 \mathrm{MHz}) 6.84$ $(1 \mathrm{H}, \mathrm{d}, J 2.0, \operatorname{Ar}-H), 6.73(1 \mathrm{H}, \mathrm{d}, J 2.0, \operatorname{Ar}-H), 6.31(1 \mathrm{H}, \mathrm{d}$, $\left.J 15.7,1^{\prime}-\mathrm{H}\right), 6.16\left(1 \mathrm{H}, \mathrm{dt}, J 15.7,6.8,2^{\prime}-\mathrm{H}\right), 5.26(2 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{OCH}_{2} \mathrm{O}\right), 5.23\left(2 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{2} \mathrm{O}\right), 3.88\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{CH}_{3}\right), 3.84$ $3.76\left(5 \mathrm{H}, \mathrm{m}, 10^{\prime}-\mathrm{H}, 2 \times \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 3.66-3.58\left(1 \mathrm{H}, \mathrm{m}, 6^{\prime}-\mathrm{H}\right)$,3.57-3.51 ( $4 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}$ ), $3.37\left(3 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{3}\right), 3.36$ $\left(3 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{3}\right), 2.25-2.14\left(2 \mathrm{H}, \mathrm{m}, 2 \times 3^{\prime}-\mathrm{H}\right), 1.75-1.37(10 \mathrm{H}, \mathrm{m}$, $\left.2 \times 4^{\prime}-\mathrm{H}, 2 \times 5^{\prime}-\mathrm{H}, 2 \times 7^{\prime}-\mathrm{H}, 2 \times 8^{\prime}-\mathrm{H}, 2 \times 9^{\prime}-\mathrm{H}\right), 1.18(3 \mathrm{H}, \mathrm{d}$, $\left.J 6.2,3 \times 11^{\prime}-\mathrm{H}\right) ; \delta_{\mathrm{C}}(100 \mathrm{MHz}) 168.5\left(\mathrm{CO}_{2} \mathrm{CH}_{3}\right), 158.8$, (quat. Ar-C), 155.2 (quat. Ar-C), 137.8 (quat. Ar-C), 134.4 ( $2^{\prime}-\mathrm{C}$ ), 126.5 ( $\left.1^{\prime}-\mathrm{C}\right), 117.2$ (quat. Ar-C), 105.9 (Ar-C), 102.5 (Ar-C), $93.8\left(\mathrm{OCH}_{2} \mathrm{O}\right), 93.2\left(\mathrm{OCH}_{2} \mathrm{O}\right), 71.5\left(6{ }^{\prime}-\mathrm{C}, 2 \times \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right)$, $67.9\left(10^{\prime}-\mathrm{C}\right), 67.8\left(\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 67.7\left(\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 59.0$ $\left(2 \times \mathrm{OCH}_{3}\right), 52.3\left(\mathrm{CO}_{2} \mathrm{CH}_{3}\right), 39.1\left(\mathrm{CH}_{2}\right), 37.2\left(\mathrm{CH}_{2}\right), 36.9$ $\left(\mathrm{CH}_{2}\right), 33.0\left(3^{\prime}-\mathrm{C}\right), 25.0\left(\mathrm{CH}_{2}\right), 23.6\left(11^{\prime}-\mathrm{C}\right), 21.8\left(\mathrm{CH}_{2}\right) ; \mathrm{m} / \mathrm{z}$ (ES) $551\left(\mathrm{MNa}^{+}, 100 \%\right), 463$ (4), 242 (8) [Found ( $\mathrm{MNa}^{+}$) $551.2818 . \mathrm{C}_{27} \mathrm{H}_{44} \mathrm{NaO}_{10}$ requires $\mathrm{MNa}, 551.2832$ ].

## General procedure for the hydrolysis of methyl benzoates 17a and 17b

For a 0.08 mmol scale reaction: 10 M aqueous KOH solution ( 1 ml ) was added to a solution of the methyl benzoate ( 1.0 equiv.) in ethane-1,2-diol ( 2 ml ) and the reaction was heated to $120^{\circ} \mathrm{C}$ for 3 h . The reaction mixture was poured into water $(10 \mathrm{ml})$ and extracted with $\mathrm{Et}_{2} \mathrm{O}(2 \times 5 \mathrm{ml})$. The aqueous phase was acidified to pH 2 using 3 M aqueous HCl solution and extracted with $\mathrm{CHCl}_{3}(3 \times 10 \mathrm{ml})$. The combined $\mathrm{CHCl}_{3}$ extracts were washed with brine ( 2 ml ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated under reduced pressure to afford the seco acid, which was used without further purification.

## 4,6-Bis[(2-methoxyethoxy)methyloxy]-2-[(1' $\left.E, 6^{\prime} R, 10^{\prime} S\right)$ -

 $6^{\prime}, 10^{\prime}$-dihydroxyundec-1'-en-1'-yllbenzoic acid (18a). Prepared according to the general procedure using methyl benzoate 17 a $(48 \mathrm{mg}, 0.09 \mathrm{mmol})$. The seco acid 18a ( $40 \mathrm{mg}, 87 \%$ ) was obtained as a colourless liquid; $v_{\max }($ film $) / \mathrm{cm}^{-1} 3396$ (OH), 2930, 1718 (C=O), 1601 (C=C), 1578 (C=C), 1456, 1284, 1172, $1158,1109,1020,848 ; \delta_{\mathrm{H}}(400 \mathrm{MHz}) 6.82(1 \mathrm{H}, \mathrm{d}, J 1.8, \mathrm{Ar}-H)$, $6.73(1 \mathrm{H}, \mathrm{d}, J 1.8, \mathrm{Ar}-H), 6.55\left(1 \mathrm{H}, \mathrm{d}, 15.7,1^{\prime}-\mathrm{H}\right), 6.11(1 \mathrm{H}, \mathrm{dt}$, $\left.J 15.7,6.9,2^{\prime}-\mathrm{H}\right), 5.25\left(4 \mathrm{H}, \mathrm{s}, 2 \times \mathrm{OCH}_{2} \mathrm{O}\right), 4.31(2 \mathrm{H}$, br s, $2 \times \mathrm{OH}), 3.91-3.69\left(5 \mathrm{H}, \mathrm{m}, 10^{\prime}-\mathrm{H}, 2 \times \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 3.68-3.59$ $\left(1 \mathrm{H}, \mathrm{m}, 6^{\prime}-\mathrm{H}\right), 3.58-3.48\left(4 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 3.37(3 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{OCH}_{3}\right), 3.32\left(3 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{3}\right), 2.26-2.11\left(2 \mathrm{H}, \mathrm{m}, 2 \times 3^{\prime}-\mathrm{H}\right), 1.66-$ $1.29\left(10 \mathrm{H}, \mathrm{m}, 2 \times 4^{\prime}-\mathrm{H}, 2 \times 5^{\prime}-\mathrm{H}, 2 \times 7^{\prime}-\mathrm{H}, 2 \times 8^{\prime}-\mathrm{H}, 2 \times 9^{\prime}-\mathrm{H}\right)$, $1.18\left(3 \mathrm{H}, \mathrm{d}, J 6.1,3 \times 11^{\prime}-\mathrm{H}\right) ; \delta_{\mathrm{C}}(100 \mathrm{MHz}) 169.3\left(\mathrm{CO}_{2} \mathrm{H}\right)$, 158.9 (quat. Ar-C), 155.2 (quat. Ar-C), 138.7 (quat. Ar-C), 134.2 ( $2^{\prime}-\mathrm{C}$ ), 127.6 ( $1^{\prime}-\mathrm{C}$ ), 117.0 (quat. Ar-C), 106.8 (Ar-C), $102.4(\mathrm{Ar}-\mathrm{C}), 93.8\left(\mathrm{OCH}_{2} \mathrm{O}\right), 93.3\left(\mathrm{OCH}_{2} \mathrm{O}\right), 71.6\left(\mathrm{OCH}_{2}-\right.$ $\left.\mathrm{CH}_{2} \mathrm{O}\right), 71.5\left(\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right)$, 71.3 ( $\left.6^{\prime}-\mathrm{C}\right)$, 68.1 ( $\left.10^{\prime}-\mathrm{C}\right), 68.1$ $\left(\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 67.7\left(\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 59.0\left(\mathrm{OCH}_{3}\right), 58.9\left(\mathrm{OCH}_{3}\right)$, $38.8\left(\mathrm{CH}_{2}\right)$, $37.1\left(\mathrm{CH}_{2}\right), 35.6\left(\mathrm{CH}_{2}\right), 32.2\left(3^{\prime}-\mathrm{C}\right), 24.2\left(\mathrm{CH}_{2}\right)$, 23.3 (11'-C), $21.4\left(\mathrm{CH}_{2}\right)$; m/z (ES) $537\left(\mathrm{MNa}^{+}, 20 \%\right)$, 413 (59), 365 (100), 349 (79) [Found ( $\mathrm{MNa}^{+}$) 537.2709. $\mathrm{C}_{26} \mathrm{H}_{42} \mathrm{NaO}_{10}$ requires $\mathrm{MNa}, 537.2676]$.[^0]$38.8\left(\mathrm{CH}_{2}\right), 37.0\left(\mathrm{CH}_{2}\right), 35.7\left(\mathrm{CH}_{2}\right), 32.2\left(3^{\prime}-\mathrm{C}\right), 24.1\left(\mathrm{CH}_{2}\right)$, 23.5 (11'-C), $21.5\left(\mathrm{CH}_{2}\right) ; \mathrm{m} / \mathrm{z}(\mathrm{ES}) 537\left(\mathrm{MNa}^{+}, 100 \%\right), 515$ $\left(\mathrm{MH}^{+}, 17\right), 421$ (8), 242 (9) [Found ( $\mathrm{MNa}^{+}$) 537.2700. $\mathrm{C}_{26} \mathrm{H}_{42} \mathrm{NaO}_{10}$ requires $\left.\mathrm{MNa}, 537.2676\right]$.

## General procedure for the macrolactonisation of seco acids 18a and 18b using the Mukaiyama protocol

For a 0.10 mmol scale reaction: a solution of the seco acid (1.0 equiv.) and $\mathrm{Et}_{3} \mathrm{~N}$ (8.0 equiv.) in $\mathrm{MeCN}(10 \mathrm{ml})$ was added to a refluxing solution of 2-chloro-1-methylpyridinium iodide (4.0 equiv.) in $\mathrm{MeCN}(10 \mathrm{ml})$ over a 10 h period via syringe pump. After an additional 5 h at reflux, the solvent was removed under reduced pressure. Purification of the residue by flash column chromatography (silica gel, $3 \% \mathrm{MeOH}$ in $\mathrm{Et}_{2} \mathrm{O}$ ) afforded the diprotected zearalenol.
2,4-Bis[(2-methoxyethoxy)methyl]-a-zearalenol (19a). Prepared according to the general procedure using seco acid 18a ( $50 \mathrm{mg}, 0.10 \mathrm{mmol}$ ). The diprotected $\alpha$-zearalenol 19 a ( 32 mg , $64 \%$ ) was obtained as a colourless oil; $[a]_{\mathrm{D}}^{30}+43.4$ (c 1.1 in $\left.\mathrm{CHCl}_{3}\right) ; v_{\text {max }}($ film $) / \mathrm{cm}^{-1} 3442(\mathrm{OH}), 2921,1719$ (C=O), 1600 (C=C), 1577 (C=C), 1453, 1261, 1155, 1104, 1022; $\delta_{\mathrm{H}}(400 \mathrm{MHz})$ $6.82(1 \mathrm{H}, \mathrm{d}, J 2.0, \mathrm{Ar}-H), 6.73(1 \mathrm{H}, \mathrm{d}, J 2.0, \mathrm{Ar}-H), 6.46(1 \mathrm{H}$, d, J 16.1, $\left.1^{\prime}-\mathrm{H}\right), 6.22\left(1 \mathrm{H}, \mathrm{dt}, J 16.1,6.0,2^{\prime}-\mathrm{H}\right), 5.30-5.19(5 \mathrm{H}$, $\left.\mathrm{m}, 10^{\prime}-\mathrm{H}, 2 \times \mathrm{OCH}_{2} \mathrm{O}\right), 3.83-3.77\left(4 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right)$, $3.77-3.69\left(1 \mathrm{H}, \mathrm{m}, 6^{\prime}-\mathrm{H}\right), 3.59-3.51\left(4 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right)$, $3.38\left(3 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{3}\right), 3.37\left(3 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{3}\right), 2.38-2.28(1 \mathrm{H}, \mathrm{m}$, $\left.3^{\prime}-\mathrm{H}\right), 2.24-2.12\left(1 \mathrm{H}, \mathrm{m}, 3^{\prime}-\mathrm{H}\right), 1.79-1.14$ ( $13 \mathrm{H}, \mathrm{m}$ [incl. 1.34, $\left.3 \mathrm{H}, \mathrm{d}, J 6.3,3 \times 11^{\prime}-\mathrm{H}\right] 2 \times 4^{\prime}-\mathrm{H}, 2 \times 5^{\prime}-\mathrm{H}, 2 \times 7^{\prime}-\mathrm{H}, 2 \times 8^{\prime}-\mathrm{H}$, $\left.2 \times 9^{\prime}-\mathrm{H}, 3 \times 11^{\prime}-\mathrm{H}\right) ; \delta_{\mathrm{C}}(100 \mathrm{MHz}) 167.8(\mathrm{ArCO}), 158.6$ (quat Ar-C), 154.8 (quat. Ar-C), 137.2 (quat. Ar-C), 133.8 (2'-C), 126.4 ( $1^{\prime}-\mathrm{C}$ ), 118.4 (quat. Ar-C), 106.0 (Ar-C), 102.5 (Ar-C), $93.7\left(\mathrm{OCH}_{2} \mathrm{O}\right), 93.4\left(\mathrm{OCH}_{2} \mathrm{O}\right), 71.6\left(\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 71.5$ $\left(\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right)$, 71.1 ( $6^{\prime}-\mathrm{C}$ or $10^{\prime}-\mathrm{C}$ ), 70.6 ( $10^{\prime}-\mathrm{C}$ or $6^{\prime}-\mathrm{C}$ ), 67.8 $\left(\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 67.7\left(\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 59.0\left(2 \times \mathrm{OCH}_{3}\right), 35.4$ $\left(\mathrm{CH}_{2}\right), 34.5\left(\mathrm{CH}_{2}\right), 32.5\left(\mathrm{CH}_{2}\right), 30.4\left(3^{\prime}-\mathrm{C}\right), 20.6\left(\mathrm{CH}_{2}\right), 20.4$ (11'-C), $19.9\left(\mathrm{CH}_{2}\right) ; \mathrm{m} / \mathrm{z}$ (ES) $519\left(\mathrm{MNa}^{+}, 100 \%\right)$ [Found $\left(\mathrm{MNa}^{+}\right) 519.2545 . \mathrm{C}_{26} \mathrm{H}_{40} \mathrm{NaO}_{9}$ requires $\mathrm{MNa}, 519.2570$ ].

2,4-Bis[(2-methoxyethoxy)methyl]- $\beta$-zearalenol (19b). Prepared according to the general procedure using seco acid 18b $(50 \mathrm{mg}, 0.10 \mathrm{mmol})$. The diprotected $\beta$-zearalenol $\mathbf{1 9 b}(31 \mathrm{mg}$, $62 \%$ ) was obtained as a colourless oil; $[a]_{\mathrm{D}}^{30}+40.0$ (c 0.30 in $\mathrm{CHCl}_{3}$ ); $v_{\text {max }}(\mathrm{film}) / \mathrm{cm}^{-1} 3447(\mathrm{OH}), 2926,1720(\mathrm{C}=\mathrm{O}), 1600$ (C=C), $1577(\mathrm{C}=\mathrm{C}), 1447,1264,1157,1104,1024 ; \delta_{\mathrm{H}}(600 \mathrm{MHz})$ $6.87(1 \mathrm{H}, \mathrm{d}, J 1.7, \mathrm{Ar}-H), 6.73(1 \mathrm{H}, \mathrm{d}, J 1.7, \mathrm{Ar}-H), 6.38(1 \mathrm{H}$, d, $\left.J 15.8,1^{\prime}-\mathrm{H}\right), 6.11-6.04\left(1 \mathrm{H}, \mathrm{m}, 2^{\prime}-\mathrm{H}\right), 5.41-5.33(1 \mathrm{H}, \mathrm{m}$, $\left.10^{\prime}-\mathrm{H}\right), 5.26\left(2 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{2} \mathrm{O}\right), 5.25(1 \mathrm{H}, \mathrm{d}, J 7.0, \mathrm{OC} H \mathrm{HO}), 5.22$ $(1 \mathrm{H}, \mathrm{d}, J 7.0, \mathrm{OCHHO}), 3.83-3.78\left(4 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right)$, $3.71-3.64\left(1 \mathrm{H}, \mathrm{m}, 6^{\prime}-\mathrm{H}\right), 3.53-3.48\left(4 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right)$, $3.39\left(3 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{3}\right), 3.37\left(3 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{3}\right), 2.34-2.27(1 \mathrm{H}, \mathrm{m}$, $\left.3^{\prime}-\mathrm{H}\right), 2.26-2.18\left(1 \mathrm{H}, \mathrm{m}, 3^{\prime}-\mathrm{H}\right), 1.83-1.19(13 \mathrm{H}, \mathrm{m}$ [incl. 1.30, $\left.3 \mathrm{H}, \mathrm{d}, J 6.4,3 \times 11^{\prime}-\mathrm{H}\right] 2 \times 4^{\prime}-\mathrm{H}, 2 \times 5^{\prime}-\mathrm{H}, 2 \times 7^{\prime}-\mathrm{H}, 2 \times 8^{\prime}-\mathrm{H}$, $\left.2 \times 9^{\prime}-\mathrm{H}, 3 \times 11^{\prime}-\mathrm{H}\right) ; \delta_{\mathrm{C}}(100 \mathrm{MHz}) 167.6(\mathrm{ArCO}), 158.6$ (quat. Ar-C), 154.6 (quat. Ar-C), 136.9 (quat. Ar-C), 133.8 (2'-C), 127.6 ( $1^{\prime}-\mathrm{C}$ ), 118.5 (quat. Ar-C), 105.6 (Ar-C), 102.5 (Ar-C), $93.6\left(\mathrm{OCH}_{2} \mathrm{O}\right), 93.4\left(\mathrm{OCH}_{2} \mathrm{O}\right), 71.6\left(\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 71.5$ $\left(\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 70.8\left(10^{\prime}-\mathrm{C}\right), 69.1\left(6^{\prime}-\mathrm{C}\right), 67.8\left(\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right)$, $67.7\left(\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 59.0\left(2 \times \mathrm{OCH}_{3}\right), 36.7\left(\mathrm{CH}_{2}\right), 34.9\left(\mathrm{CH}_{2}\right)$, $32.7\left(\mathrm{CH}_{2}\right), 30.3\left(3^{\prime}-\mathrm{C}\right), 22.8\left(\mathrm{CH}_{2}\right), 19.5\left(11^{\prime}-\mathrm{C}\right), 19.4\left(\mathrm{CH}_{2}\right)$; $\mathrm{m} / \mathrm{z}$ (ES) $519\left(\mathrm{MNa}^{+}, 100 \%\right), 421$ (13), 281 (37), 186 (18) [Found $\left(\mathrm{MNa}^{+}\right) 519.2605 . \mathrm{C}_{26} \mathrm{H}_{40} \mathrm{NaO}_{9}$ requires $\mathrm{MNa}, 519.2570$ ].

## General procedure for the preparation of $\alpha$ - and $\boldsymbol{\beta}$-zearalenol 1 and 2

For a 0.04 mmol scale reaction: THF ( 2 ml ) and aqueous 1.5 M HCl solution ( 1 ml ) were added to the diprotected zearalenol (1.0 equiv.) and the mixture was stirred at $40^{\circ} \mathrm{C}$ for 48 h . Aqueous $\mathrm{NaHCO}_{3}$ solution ( 2 ml ) was added and the mixture was extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 5 \mathrm{ml})$. The combined organic extracts
were dried over $\mathrm{MgSO}_{4}$ and concentrated under reduced pressure. Purification of the residue using a Biotage FLASH 12i system (FLASH 12S cartridge, $1 \% \mathrm{MeOH}$ in DCM) afforded the natural product zearalenol.
$\alpha$-Zearalenol (1). Prepared according to the general procedure using diprotected $\alpha$-zearalenol 19a ( $20 \mathrm{mg}, 0.04 \mathrm{mmol}$ ). $\alpha$-Zearalenol $\mathbf{1}$ ( $12 \mathrm{mg}, 93 \%$ ) was obtained as a colourless oil. $600 \mathrm{MHz}{ }^{1} \mathrm{H}$ NMR analysis indicated a de of $94 \%$; $[a]_{\mathrm{D}}^{32}-93.6$ (c 0.55 in acetone) [optical rotation obtained on an authentic sample: $[a]_{\mathrm{D}}^{32}-97.3$ (c 0.55 in acetone)]; $v_{\text {max }}($ film $) / \mathrm{cm}^{-1} 3362$ ( OH ), 2933, 2859, 1644 (C=O), 1608 ( $\mathrm{C}=\mathrm{C}$ ), 1579 ( $\mathrm{C}=\mathrm{C}$ ), 1454, 1391, 1354, 1314, 1258, 1204, 1162, 1110, 1026, 969, 910, 848, $812 ; \delta_{\mathrm{H}}\left(600 \mathrm{MHz}, \mathrm{d}_{6}\right.$-acetone) $12.17(1 \mathrm{H}, \mathrm{s}, 2-\mathrm{OH}), 9.12(1 \mathrm{H}$, br s, $4-\mathrm{OH}), 7.17\left(1 \mathrm{H}, \mathrm{d}, J 15.1,1^{\prime}-\mathrm{H}\right), 6.45(1 \mathrm{H}, \mathrm{d}, J 2.4$, Ar- $H$ ), $6.29(1 \mathrm{H}, \mathrm{d}, J 2.4, \operatorname{Ar}-H), 5.76-5.69\left(1 \mathrm{H}, \mathrm{m}, 2^{\prime}-\mathrm{H}\right)$, $5.00-4.93\left(1 \mathrm{H}, \mathrm{m}, 10^{\prime}-\mathrm{H}\right), 3.80-3.72\left(1 \mathrm{H}, \mathrm{m}, 6^{\prime}-\mathrm{H}\right), 3.31(1 \mathrm{H}, \mathrm{br}$ $\left.\mathrm{s}, 6^{\prime}-\mathrm{OH}\right), 2.34-2.27\left(2 \mathrm{H}, \mathrm{m}, 2 \times 3^{\prime}-\mathrm{H}\right), 1.96-1.87\left(2 \mathrm{H}, \mathrm{m}, 4^{\prime}-\mathrm{H}\right.$, $\left.9^{\prime}-\mathrm{H}\right), 1.72-1.41\left(7 \mathrm{H}, \mathrm{m}, 4^{\prime}-\mathrm{H}, 5^{\prime}-\mathrm{H}, 2 \times 7^{\prime}-\mathrm{H}, 2 \times 8^{\prime}-\mathrm{H}, 9^{\prime}-\mathrm{H}\right)$, $1.39\left(3 \mathrm{H}, \mathrm{d}, J 6.1,3 \times 11^{\prime}-\mathrm{H}\right), 1.12-1.08\left(1 \mathrm{H}, \mathrm{m}, 5^{\prime}-\mathrm{H}\right) ; \delta_{\mathrm{C}}(100$ $\mathrm{MHz}, \mathrm{d}_{6}$-acetone) 171.8 ( ArCO ), 165.9 (quat. Ar-C), 162.5 (quat. Ar-C), 144.3 (quat. Ar-C), 133.5 ( $\left.2^{\prime}-\mathrm{C}\right), 132.7$ ( $1^{\prime}-\mathrm{C}$ ), 108.7 (Ar-C), 102.6 (quat. Ar-C), 101.9 (Ar-C), 73.8 ( $10^{\prime}-\mathrm{C}$ ), 65.1 ( $\left.6^{\prime}-\mathrm{C}\right), 36.8$ ( $7^{\prime}-\mathrm{C}$ ), 34.8 ( $\left.9^{\prime}-\mathrm{C}\right)$, 31.8 ( $5^{\prime}-\mathrm{C}$ ), 30.3 ( $3^{\prime}-\mathrm{C}$ ), 22.6 ( $4^{\prime}-\mathrm{C}$ ), 21.5 ( $8^{\prime}-\mathrm{C}$ ), 20.3 ( $11^{\prime}-\mathrm{C}$ ); $m / z$ (ES) 343 ( $\mathrm{MNa}^{+}$, $88 \%), 325\left(\mathrm{MNa}-\mathrm{H}_{2} \mathrm{O}, 100\right), 307\left(\mathrm{MNa}-2 \mathrm{H}_{2} \mathrm{O}, 5\right), 299$ ( $\mathrm{MNa}-\mathrm{CO}_{2}, 28$ ), 229 (19), 197 (4) [Found ( $\mathrm{MNa}^{+}$) 343.1503. $\mathrm{C}_{18} \mathrm{H}_{24} \mathrm{NaO}_{5}$ requires $\left.\mathrm{MNa}, 343.1516\right]$. Data obtained on the synthetic $\alpha$-zearalenol 1 were consistent ( ${ }^{1} \mathrm{H}$ NMR, ${ }^{13} \mathrm{C}$ NMR, TLC, $[a]_{\mathrm{D}}$ ) with those obtained for an authentic sample purchased from Sigma Aldrich.
$\boldsymbol{\beta}$-Zearalenol (2). Prepared according to the general procedure using diprotected $\beta$-zearalenol 19b ( $20 \mathrm{mg}, 0.04 \mathrm{mmol}$ ). $\beta$-Zearalenol 2 ( $12 \mathrm{mg}, 93 \%$ ) was obtained as a white, amorphous solid. $600 \mathrm{MHz}{ }^{1} \mathrm{H}$ NMR analysis indicated a de of $>95 \%$; $[a]_{D}^{32}-12.5$ (c 1.00 in acetone) [optical rotation obtained on an authentic sample: $[a]_{\mathrm{D}}^{32}-12.9$ (c 1.00 in acetone) $] ; v_{\max }($ film $) /$ $\mathrm{cm}^{-1} 3396(\mathrm{OH}), 2922,2851,1643(\mathrm{C}=\mathrm{O}), 1608(\mathrm{C}=\mathrm{C}), 1580$ (C=C), 1452, 1378, 1353, 1311, 1258, 1197, 1162, 1109, 1024, $964,904,847 ; \delta_{\mathrm{H}}\left(600 \mathrm{MHz}, \mathrm{d}_{6}\right.$-acetone) $11.03(1 \mathrm{H}, \mathrm{br} \mathrm{s}, 2-\mathrm{OH})$, $9.02(1 \mathrm{H}, \mathrm{br} \mathrm{s}, 4-\mathrm{OH}), 6.86\left(1 \mathrm{H}, \mathrm{d}, J 15.5,1^{\prime}-\mathrm{H}\right), 6.53(1 \mathrm{H}, \mathrm{d}$, $J 2.2, \operatorname{Ar}-H), 6.28(1 \mathrm{H}, \mathrm{d}, J 2.2, \mathrm{Ar}-H), 5.97(1 \mathrm{H}, \mathrm{ddd}, J 15.5$, $\left.8.4,6.1,2^{\prime}-\mathrm{H}\right), 5.13-5.06\left(1 \mathrm{H}, \mathrm{m}, 10^{\prime}-\mathrm{H}\right), 3.81-3.71(1 \mathrm{H}, \mathrm{m}$, $\left.6^{\prime}-\mathrm{H}\right), 3.40\left(1 \mathrm{H}, \mathrm{br}\right.$ s, $\left.6^{\prime}-\mathrm{OH}\right), 2.36-2.28\left(1 \mathrm{H}, \mathrm{m}, 3^{\prime}-\mathrm{H}\right), 2.28-$ $2.21\left(1 \mathrm{H}, \mathrm{m}, 3^{\prime}-\mathrm{H}\right), 1.96-1.86\left(1 \mathrm{H}, \mathrm{m}, 9^{\prime}-\mathrm{H}\right), 1.79-1.61(5 \mathrm{H}, \mathrm{m}$, $\left.2 \times 4^{\prime}-\mathrm{H}, 5^{\prime}-\mathrm{H}, 7^{\prime}-\mathrm{H}, 9^{\prime}-\mathrm{H}\right), 1.59-1.52\left(1 \mathrm{H}, \mathrm{m}, 7^{\prime}-\mathrm{H}\right), 1.49-1.37$ $\left(2 \mathrm{H}, \mathrm{m}, 2 \times 8^{\prime}-\mathrm{H}\right), 1.34\left(3 \mathrm{H}, \mathrm{d}, J 6.3,3 \times 11^{\prime}-\mathrm{H}\right), 1.32-1.23(1 \mathrm{H}$, $\left.\mathrm{m}, 5^{\prime}-\mathrm{H}\right) ; \delta_{\mathrm{C}}\left(100 \mathrm{MHz}, \mathrm{d}_{6}\right.$-acetone) 171.1 ( ArCO ), 163.6 (quat. Ar-C), 161.8 (quat. Ar-C), 142.3 (quat. Ar-C), 132.2 ( $2^{\prime}-\mathrm{C}$ ), $131.4\left(1^{\prime}-\mathrm{C}\right), 107.4$ (Ar-C), 105.2 (quat. Ar-C), 101.6 (Ar-C), 73.4 ( $10^{\prime}-\mathrm{C}$ ), 67.6 ( $6^{\prime}-\mathrm{C}$ ), 36.3 ( $7^{\prime}-\mathrm{C}$ ), 34.2 ( $9^{\prime}-\mathrm{C}$ ), 31.6 ( $5^{\prime}-\mathrm{C}$ ), 31.0 ( $3^{\prime}-\mathrm{C}$ ), 22.4 ( $4^{\prime}-\mathrm{C}$ ), 19.1 ( $8^{\prime}-\mathrm{C}$ ), 18.0 ( $11^{\prime}-\mathrm{C}$ ); $m / z(\mathrm{ES}) 343$ $\left(\mathrm{MNa}^{+}, 77 \%\right), 325\left(\mathrm{MNa}-\mathrm{H}_{2} \mathrm{O}, 100\right), 299\left(\mathrm{MNa}-\mathrm{CO}_{2}\right.$, 29) [Found ( $\mathrm{MNa}^{+}$) 343.1510. $\mathrm{C}_{18} \mathrm{H}_{24} \mathrm{NaO}_{5}$ requires MNa , 343.1516]. Data obtained for the synthetic $\beta$-zearalenol 2 were consistent ( ${ }^{1} \mathrm{H}$ NMR, ${ }^{13} \mathrm{C}$ NMR, TLC, $[a]_{\mathrm{D}}$ ) with those obtained for an authentic sample purchased from Sigma Aldrich.

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## References

1 For a review see: V. Betina, in Mycotoxins: Chemical, Biological and Environmental Aspects, ed. V. Betina, Elsevier, Amsterdam, 1989, p. 271.

2 W. T. Shier, Rev. Med. Vet. (Toulouse), 1998, 149, 599.
3 W. M. Hagler, C. J. Mirocha, S. V. Pathre and J. C. Behrens, Appl. Environ. Microbiol., 1979, 37, 849.
4 K. P. Lone, CRC Crit. Rev. Food Sci. Nutr., 1997, 37, 93.
5 W. H. Utian, Br. Med. J., 1973, 1, 579.
6 For selected racemic syntheses of zearalenone see: (a) D. Taub, N. N. Girotra, R. D. Hoffsommer, C. H. Kuo, H. L. Slates, S. Weber and N. L. Wendler, Tetrahedron, 1968, 24, 2443; (b) E. J. Corey and K. C. Nicolaou, J. Am. Chem. Soc., 1974, 96, 5614; (c) S. Masamune, S. Kamata and W. Schilling, J. Am. Chem. Soc., 1975, 97, 3515; (d) T. Takahashi, H. Ikeda and J. Tsuji, Tetrahedron Lett., 1981, 22, 1363 and references therein; for enantioselective syntheses of naturally occurring $(S)$-zearalenone, see: (e) G. Solladié, M. Carmen Maestro, A. Rubio, Concepcion Pedregal, M. Carmen Carreño and José L. Garcia Ruano, J. Org. Chem., 1991, 56, 2317; ( $f$ ) A. Kalivretenos, J. K. Stille and L. S. Hegedus, J. Org. Chem., 1991, 56, 2883; (g) E. Keinan, S. C. Sinah and A. Sinah-Bagchi, J. Chem. Soc., Perkin Trans. 1, 1991, 3333; (h) S. A. Hitchcock and G. Pattenden, J. Chem. Soc., Perkin Trans. 1, 1992, 1323; (i) Z. Q. Wang and S. K. Tian, Chin. Chem. Lett., 1997, 8, 591; (j) K. C. Nicolaou, N. Wissinger, J. Pastor and F. Murphy, Angew. Chem., Int. Ed., 1998, 37, 2534; (k) A. Fürstner, O. R. Thiel, N. Kindler and B. Bartkowska, J. Org. Chem., 2000, 65, 7990 and references therein.

7 S. V. Ley and S. Burckhardt, J. Chem. Soc., Perkin Trans. 1, 2000, 3028.

8 S. V. Ley, G. Meek, K.-H. Metten and C. Piqué, J. Chem. Soc., Chem. Commun, 1994, 1931; S. V. Ley, L. R. Cox, G. Meek, K.-H. Metten, C. Piqué and J. M. Worrall, J. Chem. Soc., Perkin Trans. 1, 1997, 3299.
9 S. V. Ley, S. Burckhardt, L. R. Cox and J. M. Worrall, Chem. Commun., 1998, 229; S. V. Ley, S. Burckhardt, L. R. Cox and J. M. Worrall, J. Chem. Soc., Perkin Trans. 1, 2000, 211.

10 S. Servi, J. Org. Chem., 1985, 50, 5865.
11 P. A. Grieco and C. S. Pogonowski, J. Am. Chem. Soc., 1973, 95, 3071.

12 Y. Gao and K. B. Sharpless, J. Am. Chem. Soc., 1988, 110, 7538.
13 M. Caruso, J. G. Knight and S. V. Ley, Synlett, 1990, 224.
14 D. M. Hodgson, A. M. Foley and P. J. Lovell, Tetrahedron Lett., 1998, 39, 6419.
15 V. Farina and B. Krishnan, J. Am. Chem. Soc., 1991, 113, 9585.

16 T. Mukaiyama, R. Matsueda and M. Suzuki, Tetrahedron Lett., 1970, 1901.
17 J. Inanaga, K. Hirata, H. Saeki, T. Katsuki and M. Yamaguchi, Bull. Chem. Soc. Jpn., 1979, 52, 1989.
18 T. Mukaiyama, M. Usui and K. Saigo, Chem. Lett., 1976, 49.


[^0]:    4,6-Bis[(2-methoxyethoxy)methyloxy]-2-[(1' $\left.E, 6^{\prime} S, 10^{\prime} S\right)$ $6^{\prime}, \mathbf{1 0}$ '-dihydroxyundec-1'-en-1'-yl]benzoic acid (18b). Prepared according to the general procedure using methyl benzoate 17b $(37 \mathrm{mg}, 0.07 \mathrm{mmol})$. The seco acid 18b ( $33 \mathrm{mg}, 91 \%$ ) was obtained as a colourless liquid; $v_{\max }($ film $) / \mathrm{cm}^{-1} 3376(\mathrm{OH})$, 2932, 1720 (C=O), 1601 (C=C), 1579 (C=C), 1455, 1317, 1285, $1174,1158,1111,1021,849 ; \delta_{\mathrm{H}}(600 \mathrm{MHz}) 6.82(1 \mathrm{H}, \mathrm{d}, J 1.9$, Ar-H), 6.73 ( $1 \mathrm{H}, \mathrm{d}, J 1.9, \operatorname{Ar}-H$ ), $6.58\left(1 \mathrm{H}, \mathrm{d}, 15.7,1^{\prime}-\mathrm{H}\right), 6.12$ $\left(1 \mathrm{H}, \mathrm{dt}, J 15.7,6.9,2^{\prime}-\mathrm{H}\right), 5.27\left(2 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{2} \mathrm{O}\right), 5.26(2 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{OCH}_{2} \mathrm{O}\right), 3.88-3.70\left(5 \mathrm{H}, \mathrm{m}, 10^{\prime}-\mathrm{H}, 2 \times \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 3.68-3.61$ $\left(1 \mathrm{H}, \mathrm{m}, 6^{\prime}-\mathrm{H}\right), 3.58-3.51\left(4 \mathrm{H}, \mathrm{m}, 2 \times \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 3.37(3 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{OCH}_{3}\right), 3.32\left(3 \mathrm{H}, \mathrm{s}, \mathrm{OCH}_{3}\right), 2.26-2.18\left(2 \mathrm{H}, \mathrm{m}, 2 \times 3^{\prime}-\mathrm{H}\right), 1.69-$ $1.39\left(10 \mathrm{H}, \mathrm{m}, 2 \times 4^{\prime}-\mathrm{H}, 2 \times 5^{\prime}-\mathrm{H}, 2 \times 7^{\prime}-\mathrm{H}, 2 \times 8^{\prime}-\mathrm{H}, 2 \times 9^{\prime}-\mathrm{H}\right)$, $1.18\left(3 \mathrm{H}, \mathrm{d}, J 6.2,3 \times 11^{\prime}-\mathrm{H}\right) ; \delta_{\mathrm{C}}(150 \mathrm{MHz}) 169.1\left(\mathrm{CO}_{2} \mathrm{H}\right)$, 158.9 (quat. Ar-C), 155.2 (quat. Ar-C), 139.0 (quat. Ar-C), 134.2 ( $2^{\prime}-\mathrm{C}$ ), 127.7 ( $1^{\prime}-\mathrm{C}$ ), 116.8 (quat. Ar-C), 106.9 (Ar-C), 102.3 (Ar-C), $93.7\left(\mathrm{OCH}_{2} \mathrm{O}\right), 93.3\left(\mathrm{OCH}_{2} \mathrm{O}\right), 71.6\left(\mathrm{OCH}_{2}-\right.$ $\left.\mathrm{CH}_{2} \mathrm{O}\right), 71.5\left(\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 71.3\left(6^{\prime}-\mathrm{C}\right), 68.2$ ( $\left.10^{\prime}-\mathrm{C}\right), 68.2$ $\left(\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 67.7\left(\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right), 59.0\left(\mathrm{OCH}_{3}\right), 58.9\left(\mathrm{OCH}_{3}\right)$,

